

## Authority

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# Mathematics Standard 2 Year 12

## Networks Topic Guide

*The Mathematics syllabuses are the documents used to inform the scope of content that will be assessed in the HSC examinations.*

*Topic Guides provide support for the Mathematics Stage 6 courses. They contain information organised under the following headings: Terminology; Use of technology; Background information; General comments; Future study; Considerations and teaching strategies; Suggested applications and exemplar questions.*

*Topic Guides illustrate ways to explore syllabus-related content and consequently do not define the scope of problems or learning experiences that students may encounter through their study of a topic. The terminology list contains terms that may be used in the teaching and learning of the topic. The list is not exhaustive and is provided simply to aid discussion.*

*Please provide any feedback to the Mathematics and Numeracy Curriculum Inspector.*

<b>Effective from</b>	2018 Year 11, and Term 4, 2018 Year 12
<b>Publication date</b>	October 2018
<b>Updated</b>	December 2018 – new NESA template implemented; clarification of the LFT definition, correction to a solution

# Guide To Networks

**B Lingard**



## Guide To Networks:

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web the pcl<sub>3</sub> lewis structure has the typical case of phosphorus p in the center with 3 bonds to 3 other atoms phosphorus is from the same column as nitrogen in the periodic table meaning that p and n generally have the same bonding structure note

the lone pair dots without bonds on top of p just like for n in the previous example for nh<sub>3</sub> chlorine cl

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web molecular models and 3d printing activity lewis dot structures and molecule geometries worksheet answer key 6 10 co 3 2 total number of valence electrons 24 cad engineered 3d sketch model show dipole arrows lewis structure is there a polar bond in this molecule yes or no vsepr shape name trigonal planar bond angles 120

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web jun 8 2022 lewis structure in chemistry lewis structures also called electron dot structures or electron dot diagrams are diagrams that show the bonding between atoms of a molecule and the lone pairs of

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web this exercise required us to take the formulas in the data table and calculate the number of valence electrons using the periodic table draw its lewis structure on paper build the molecular model and record a photo list the number of bonds and lone pairs surrounding the central atom and lastly identify the structure geometry