



# Gizmos

Name: \_\_\_\_\_ Date: \_\_\_\_\_

## Student Exploration: pH Analysis

**Vocabulary** acid, acids, alkaline, base, indicator, neutral, pH

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo. [5 marks])

- Acids** are substances that produce hydrogen ions ( $H^+$ ) when dissolved in water. Lemon juice is an example of an acid.
  - What does lemon juice taste like? \_\_\_\_\_
  - What does it feel like if lemon juice gets in your eye? \_\_\_\_\_
- Bases** are substances that produce hydroxide ions ( $OH^-$ ) when dissolved in water. Hand soap is an example of a base.
  - What does soap feel like? \_\_\_\_\_
  - What does soap taste like? \_\_\_\_\_
  - What does it feel like if soap gets in your eye? \_\_\_\_\_

### Gizmo Warmup (3 marks)

The strength of an acid or base is measured on the **pH** scale. The term "pH" is short for "potential of hydrogen." It is a measure of how many excess  $H^+$  ions there are in a solution. The pH scale runs from 0 to 14, with 7 representing the highest concentration of hydrogen ions. **Acidic** substances have a pH below 7, while **alkaline** substances (bases) have a pH above 7. Pure water has



# Explore Learning Ph Analysis Gizmo Answer Key

**Hussin A.Rothana**



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