



- 1 The photoelectric effect is the emission of electrons by light.  
Threshold at  $\nu_0 \approx 5 \cdot 10^4$  Hz (caesium:  $\nu_0 = 5.5 \cdot 10^4$  Hz).
- 2 Hertz first observed the effect in 1887.  
Einstein explained it in 1905:  $E = h \cdot f$  ( $h = 6.626 \cdot 10^{-34}$  Js).
- 3 The work function  $\Phi$  is the escape energy of the electrons.  
Metal values are between 2–5 eV.
- 4 Limit frequency  $\nu_0 = \Phi / h$  for electron escape.  
At  $\Phi = 4$  eV,  $\nu_0 \approx 9.7 \cdot 10^4$  Hz.
- 5 Photon energy  $E = h \cdot f$  in joules.  
For  $f = 6 \cdot 10^4$  Hz,  $E \approx 4 \cdot 10^{-19}$  J ( $\approx 2.5$  eV).
- 6 A light quantum is a photon with energy  $E = h \cdot f$ .  
Laser ( $\lambda = 632$  nm) emits  $\approx 3 \cdot 10^5$  photons/s·mW.
- 7 Current only flows at  $f > \nu_0$ .  
 $I$  up to 10  $\mu$ A at  $I_0 = 1$  mW/cm<sup>2</sup>.
- 8 The counterfield method determines the maximum electron energy.  
 $E_{\text{max}} = e \cdot U_{\text{c}}$ ; at  $U_{\text{c}} = 1$  V,  $E_{\text{max}} = 1$  eV.
- 9 Photomultipliers multiply photoelectrons.  
Amplification up to  $10^6$  at  $-1000$  V operation.
- 10 Solar cells use the photoelectric effect to generate electricity.  
Efficiency  $\approx 20$  %, voltage  $\approx 0.6$  V per cell.

# Explorelearning Photoelectric Effect Activity Answers

**M Tight**



**Explorelearning Photoelectric Effect Activity Answers:**

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