



Name: \_\_\_\_\_  
 Period: \_\_\_\_\_

KEY

## Nuclear Radiation Worksheet

1. a. What determines whether an atom is stable or unstable?

*Its overall size and a difference between proton # and neutron #*

b. How does an unstable atom gain stability?

*It undergoes radioactive decay*

c. What part of an atom undergoes change during radioactive decay?

*The nucleus!*

2. Which of the human senses can detect radiation?

*Our eyes can detect colors, but we have no senses to detect alpha, beta, or gamma radiation*

3. Explain the difference between an isotope and a radioactive isotope.

*An isotope has a different # of protons & neutrons. Radioactive isotope is larger, unstable, and has a different # of p*

4. Distinguish between alpha, beta, and gamma radiation on the basis of the following:

Particle	Mass	Charge	Penetrating power
Alpha	<i>4 amu, 2 protons, 2 neutrons</i>	Positive (+)	<i>Stopped by paper or skin</i>
Beta	<i>0 Neutrons (electron)</i>	Negative (-)	<i>Stopped by sheets of tin foil or thick paper</i>
Gamma	<i>0 Neutrons</i>	Neutral	<i>Stopped by several inches of lead or feet of concrete</i>

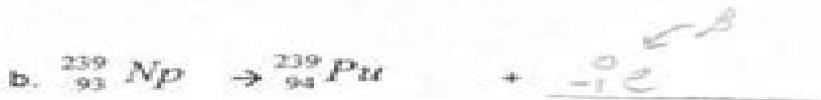
5. How are the mass number and atomic number of a nucleus affected by the loss of the following?

Particle	Mass Number Change	Atomic Number Change
Alpha	<i>-4</i>	<i>-2</i>
Beta	<i>No change</i>	<i>+1</i>
Gamma	<i>No change</i>	<i>No change</i>

6. Complete the equations for the decay processes below by supplying the missing atomic numbers and mass numbers.



7. Complete the following nuclear equations:



# Explore Learning Nuclear Decay Answer Sheet

**CL Gary**



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