

Student Exploration: Phase Changes

Approximate time: 45 minutes

Pages: 13 (with Student Answer Key)

Version: 1.0 (2014)

Prerequisite Knowledge: Students should have completed the following activities:

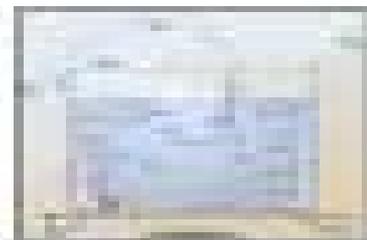
- 1. [Energy Transfer: Conduction](#) (for understanding of heat transfer) and [Energy Transfer: Convection](#) (for understanding of heat transfer).

Introduction: In this activity, you will explore the phase changes of water.

Learning Objectives:

By the end of this activity, you should be able to:

- 1. Explain the phase changes of water.
- 2. Explain the energy transfer during phase changes.
- 3. Explain the effect of temperature on the phase changes of water.



1. What is the boiling point of water?

Answer: 100°C (212°F) at sea level.

2. What is the freezing point of water?

Answer: 0°C (32°F) at sea level.

3. What is the melting point of ice?

Answer: 0°C (32°F) at sea level.

4. What is the boiling point of water at different altitudes?

Answer: The boiling point of water decreases as altitude increases.

| | |
|---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|--|
| <p>Key Concepts:</p> <ul style="list-style-type: none"> Phase changes occur when energy is added or removed from a substance. The temperature of a substance remains constant during a phase change. The amount of energy required for a phase change is proportional to the mass of the substance. | |
|---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|--|

Procedure: Use the simulation to explore the phase changes of water.

1. **Boiling:** Set the temperature to 100°C. Add heat to the water. Observe the phase change.

- a. What is the boiling point of water? **100°C** (212°F) This is the **boiling point**.
- b. What is the boiling point of water at different altitudes? **100°C** (212°F) This is the **boiling point**.
- c. What is the boiling point of water at different altitudes? **100°C** (212°F) This is the **boiling point**.

2. **Freezing:** Set the temperature to 0°C. Remove heat from the water. Observe the phase change.

3. **Melting:** Set the temperature to 0°C. Add heat to the ice. Observe the phase change.



Explore Learning Vocabulary Phase Changes Answer Key

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Explore Learning Vocabulary Phase Changes Answer Key:

The Enigmatic Realm of **Explore Learning Vocabulary Phase Changes Answer Key**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Explore Learning Vocabulary Phase Changes Answer Key** a literary masterpiece penned by way of a renowned author, readers embark on a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of people who partake in its reading experience.

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