

- How does a polar bond differ from a nonpolar bond?
- Complete the table by predicting the type of bond that will form between the following elements only using the periodic table.

Elements	Bond Type- Ionic or Covalent?
Beryllium and Fluorine	
Selenium and Chlorine	
Strontium and Sulfur	
Potassium and Iodine	
Sulfur and Iodine	
Magnesium and Chlorine	

- For each of the following molecules, determine if it is covalent, polar covalent, or ionic. Show your work by listing the electronegativities of each element in the bond.

Molecule	Electronegativity Values	Difference in Electronegativity	Bond Type? Ionic, Polar Covalent, Nonpolar Covalent?
H - Cl	H: Cl:		
H - H	H: H:		
Cl - Cl	Cl: Cl:		
C - O	C: O:		
Ca - O	Ca: O:		

Electronegativity And Polarity Study Guide Answers

SA Adler



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