

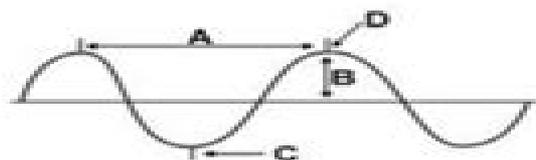
**True/False**

Indicate whether the statement is true or false.

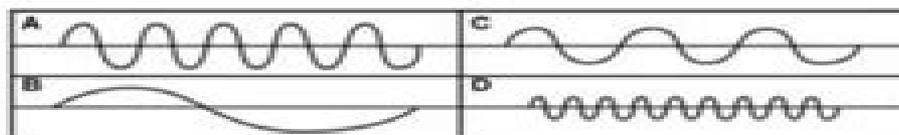
- \_\_\_ 1. There are a total of 5 *d*-orbitals.
- \_\_\_ 2. The photoelectric effect describes the emission of electrons from a metal's surface when light at a specific frequency shines on it.
- \_\_\_ 3. Although the precise location of an electron cannot be determined, it is possible to determine the probability that an electron will occupy a certain region around the nucleus.
- \_\_\_ 4. In the visible light spectrum, red light has a greater frequency (energy) than violet light.
- \_\_\_ 5.  $[\text{Kr}]5s^24d^{10}5p^5$  is the correct electron configuration for the element Bromine.
- \_\_\_ 6. The maximum number of allowed valence electrons in any atom is 8.
- \_\_\_ 7. Each element has its own unique emission spectrum.

**Multiple Choice**

Identify the choice that best completes the statement or answers the question.



- \_\_\_ 8. Which label identifies the amplitude of the wave shown?
  - a. A
  - b. B
  - c. C
  - d. D
- \_\_\_ 9. Which label identifies the wavelength of the wave shown?
  - a. A
  - b. B
  - c. C
  - d. D
- \_\_\_ 10. Which label identifies the trough of the wave shown?
  - a. A
  - b. B
  - c. C
  - d. D



- \_\_\_ 11. Which diagram shows a wave with the highest frequency?

# Electrons In Atoms Test A Answer Key

**Chemical Society (Great Britain)**



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