

Experiment 5C: Types of Chemical Reactions

Objectives:

- To observe different chemical reactions
- To interpret and explain observations and balanced chemical equations
- To identify reactions from the correct reaction type.

Materials: Please refer to Experiment 5C on page 7 of the Student Laboratory Manual.

Pre-lab Questions:

Reaction 1:

1. What are safety equipment such as goggles and aprons?
2. How do observations, before, during, and after each reaction determine chemical reactions are taking place?

Reaction 2:

1. Explain the color change that took place.
2. Why is the color change taking place? (Explain the effects copper has on the surface part of the substance it has reacted)

Reaction 3:

1. What color did you observe? (What color was the surface of the lead looking?)
2. How did the lead react with silver nitrate solution? (What color did you see? What color was the solution?)
3. How approximately, if possible, would the lead react with any changes in solution lead and silver nitrate? (What is the reaction product looking?)

Reaction 4:

1. How does lead copper (2) sulfate react with it? (What color did you see?)
2. Why is the color change taking place? (Explain the effects copper has on the surface part of the substance it has reacted. How did the lead react? (What color did you see?) (What color was the solution?)
3. How approximately, if possible, would the lead react with any changes in solution lead and copper sulfate? (What is the reaction product looking?)

Lab Experiment 5c Answers

JG Myers



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