

Unit 6 – Worksheet 3

Ionic Compounds

Properties

Basic structural unit

1. Give the name of the following simple binary ionic compounds.

- a. Na_2O
- b. K_2S
- c. MgCl_2
- d. CaBr_2
- e. BaI_2
- f. Al_2S_3
- g. CaBr
- h. AgF

2. Give the name of the following simple binary ionic compounds.

- a. Na_3N
- b. K_2O
- c. AgBr
- d. MgI_2
- e. SrO

3. Write the formula for the following binary ionic compounds.

- a. lithium bromide
- b. sodium iodide
- c. silver sulfide
- d. cesium oxide
- d. beryllium iodide
- f. barium hydride
- g. aluminum fluoride
- h. potassium oxide

4. Write the formula for these ionic substances.

- a. silver oxide
- b. aluminum sulfide
- c. sodium nitride
- d. barium chloride
- e. strontium hydride

Modeling Chemistry Unit Ws 2

RD Boyd



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Synthesis, Modelling and Characterization of 2D Materials and their Heterostructures Eui-Hyeok Yang, Dibakar Datta, Junjun Ding, Grzegorz Hader, 2020-06-19 Synthesis Modelling and Characterization of 2D Materials and Their Heterostructures provides a detailed discussion on the multiscale computational approach surrounding atomic molecular and atomic informed continuum models In addition to a detailed theoretical description this book provides example problems sample code script and a discussion on how theoretical analysis provides insight into optimal experimental design Furthermore the book addresses the growth mechanism of these 2D materials the formation of defects and different lattice mismatch and interlayer interactions Sections cover direct band gap Raman scattering extraordinary strong light matter interaction layer dependent photoluminescence and other physical properties Explains multiscale computational techniques from atomic to continuum scale covering different time and length scales Provides fundamental theoretical insights example problems sample code and exercise problems Outlines major characterization and synthesis methods for different types of 2D materials

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