

Ionic Compounds and Formula Worksheet

Write the formula for each of the following compounds.

1. Magnesium oxide	MgO	2. Sodium fluoride	NaF
3. Aluminum nitride	Al₂N₃	4. Potassium sulfide	K₂S
5. Lithium iodide	LiI	6. Calcium bromide	CaBr₂
7. Beryllium oxide	BeO	8. Nickel chloride	NiCl
9. Magnesium nitride	Mg₃N₂	10. Aluminum sulfide	Al₂S₃
11. Calcium phosphide	Ca₃P₂	12. Zinc fluoride	ZnF
13. Sodium oxide	Na₂O	14. Lithium chloride	LiCl
15. Zinc oxide	ZnO	16. Silver chloride	AgCl
17. Barium bromide	BaBr₂	18. Potassium phosphide	K₃P
19. Silver oxide	Ag₂O		

Write the names for the following compounds.

1. Li ₂ O	Lithium oxide	2. CaF ₂	Calcium fluoride
3. AlCl ₃	Aluminum chloride	4. K ₃ P	Potassium phosphide
5. MgS	Magnesium sulfide	6. ZnF ₂	Zinc fluoride
7. CaO	Calcium oxide	8. AgCl	Silver chloride
9. KBr	Potassium bromide	10. Al ₂ S ₃	Aluminum sulfide
11. BeF	Beryllium fluoride	12. BaBr ₂	Barium bromide
13. Na ₃ N	Sodium nitride	14. CaS	Calcium sulfide
15. Al ₂ O ₃	Aluminum oxide	16. KI	Potassium iodide
17. Na ₂ S	Sodium sulfide	18. ZnO	Zinc oxide
19. Mg ₃ P ₂	Magnesium phosphide		

Formulae Of Ionic Compounds Answer Key

**Yi-Noo Tang, Wendy L. Keeney-
Kennicutt**



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