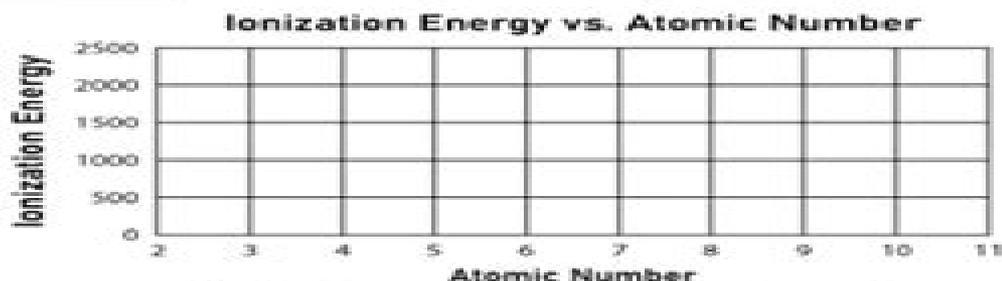


# Graphing Periodic Trends Worksheet

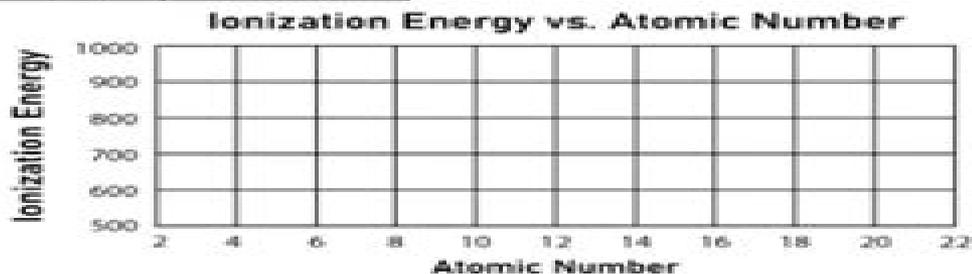
1. Using the ionization energy (IE) values of the elements in Period 2 listed below, make a line graph of IE vs. the atomic number.

Period 2					
Element	Atomic Number	IE (kJ/mol)	Element	Atomic Number	IE (kJ/mol)
Li	3	519	Li	3	519
Be	4	900	Be	4	900
B	5	799	B	5	799
C	6	1088	C	6	1088



- a) What trend do you notice for the ionization energies in Period 2?
- b) Explain why you observe this trend.
2. Make a scatter plot of ionization energy vs. atomic number for the first three atoms in Group 2A listed below.

Group 2A		
Element	Atomic Number	IE (kJ/mol)
Be	4	900
Mg	12	736
Ca	20	590



# Graphing Periodic Trends Answer Sheet

**Léo Pomar**



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## **Table of Contents Graphing Periodic Trends Answer Sheet**

1. Understanding the eBook Graphing Periodic Trends Answer Sheet
  - The Rise of Digital Reading Graphing Periodic Trends Answer Sheet
  - Advantages of eBooks Over Traditional Books
2. Identifying Graphing Periodic Trends Answer Sheet
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Graphing Periodic Trends Answer Sheet
  - User-Friendly Interface
4. Exploring eBook Recommendations from Graphing Periodic Trends Answer Sheet
  - Personalized Recommendations
  - Graphing Periodic Trends Answer Sheet User Reviews and Ratings
  - Graphing Periodic Trends Answer Sheet and Bestseller Lists
5. Accessing Graphing Periodic Trends Answer Sheet Free and Paid eBooks
  - Graphing Periodic Trends Answer Sheet Public Domain eBooks
  - Graphing Periodic Trends Answer Sheet eBook Subscription Services
  - Graphing Periodic Trends Answer Sheet Budget-Friendly Options
6. Navigating Graphing Periodic Trends Answer Sheet eBook Formats
  - ePub, PDF, MOBI, and More
  - Graphing Periodic Trends Answer Sheet Compatibility with Devices
  - Graphing Periodic Trends Answer Sheet Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Graphing Periodic Trends Answer Sheet
  - Highlighting and Note-Taking Graphing Periodic Trends Answer Sheet
  - Interactive Elements Graphing Periodic Trends Answer Sheet

8. Staying Engaged with Graphing Periodic Trends Answer Sheet
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Graphing Periodic Trends Answer Sheet
9. Balancing eBooks and Physical Books Graphing Periodic Trends Answer Sheet
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Graphing Periodic Trends Answer Sheet
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Graphing Periodic Trends Answer Sheet
  - Setting Reading Goals Graphing Periodic Trends Answer Sheet
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Graphing Periodic Trends Answer Sheet
  - Fact-Checking eBook Content of Graphing Periodic Trends Answer Sheet
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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