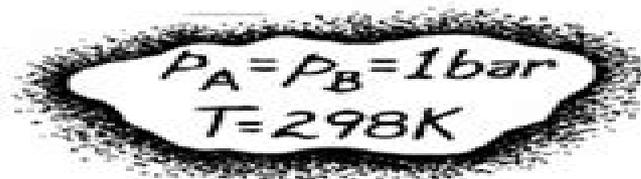


### PROBLEM 14.2

**KNOWN:** Partial pressures and temperature for a mixture of  $\text{CO}_2$  and  $\text{N}_2$ .

**FIND:** Molar concentration, mass density, mole fraction and mass fraction of each species.

**SCHEMATIC:**



$$A \rightarrow \text{CO}_2, \rho_A = 44 \text{ kg/kmol}$$

$$B \rightarrow \text{N}_2, \rho_B = 28 \text{ kg/kmol}$$

**ASSUMPTIONS:** (1) Perfect gas behavior.

**ANALYSIS:** From the equation of state for an ideal gas,

$$C_i = \frac{P_i}{RT}$$

Hence, with  $P_A = P_B$ ,

$$C_A = C_B = \frac{1 \text{ bar}}{8.314 \times 10^{-2} \text{ m}^3 \cdot \text{bar/kmol} \cdot \text{K} \times 298 \text{ K}}$$

$$C_A = C_B = 0.040 \text{ kmol/m}^3$$

With  $\rho_i = \rho_i C_i$ , it follows that

$$\rho_A = 44 \text{ kg/kmol} \times 0.04 \text{ kmol/m}^3 = 1.76 \text{ kg/m}^3$$

$$\rho_B = 28 \text{ kg/kmol} \times 0.04 \text{ kmol/m}^3 = 1.13 \text{ kg/m}^3$$

Also, with

$$x_i = C_i / \sum_j C_j$$

find

$$x_A = x_B = 0.04 / 0.08 = 0.5$$

and with

$$m_i = \rho_i / \sum_j \rho_j$$

find

$$m_A = 1.76 / (1.76 + 1.13) = 0.61$$

$$m_B = 1.13 / (1.76 + 1.13) = 0.39$$

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# Introduction To Heat Transfer 6th Edition Solution Manual Scribd

**Marcel A. Müller**



**Introduction To Heat Transfer 6th Edition Solution Manual Scribd:**

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from the outlet for about an hour to see if the tv will reset itself since the minute the tv is plugged into the wall the power supply and processor will be running 24 7 and the eprom can get corrupted by spikes

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web aug 27 2015 find the bad cap and replace it unplug the tv first then open the tv case carefully and gently gain access to the inside of the tv so you can see all the boards inside look at the capacitors on the power board for any signs of bulging

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