

CHM 1102: Advanced Assignment 11A and 11B

Name: John Brown Section #: 111

Lab section #: 10

date: 11/11/11

Assignment 11A:

| 1. | Compound | Mass | Molar Mass | Molar Mass | Mass |
|----|--------------------------------|-------|------------|------------|-------|
| | $\text{C}_2\text{H}_6\text{O}$ | 46.07 | 46.07 | 46.07 | 46.07 |
| | $\text{C}_2\text{H}_4\text{O}$ | 44.05 | 44.05 | 44.05 | 44.05 |
| | $\text{C}_2\text{H}_2\text{O}$ | 42.04 | 42.04 | 42.04 | 42.04 |

_____ 100

Assignment 11B:

| 1. net ionic eqn | 2. net ionic eqn | 3. net ionic eqn |
|---|---|---|
| $\text{Ag}^+ + \text{Cl}^- \rightarrow \text{AgCl}$ | $\text{Mg}^{2+} + 2\text{OH}^- \rightarrow \text{Mg(OH)}_2$ | $\text{Cu}^{2+} + \text{S}^{2-} \rightarrow \text{CuS}$ |

_____ 100

4. net ionic eqn = $\text{Pb}^{2+} + \text{S}^{2-} \rightarrow \text{PbS}$

_____ 100

5a. $\text{Ba}^{2+} + \text{SO}_4^{2-} \rightarrow \text{BaSO}_4$

5b. $\text{Pb}^{2+} + \text{CO}_3^{2-} \rightarrow \text{PbCO}_3$

6. net ionic eqn = $\text{Cu}^{2+} + \text{S}^{2-} \rightarrow \text{CuS}$

_____ 100

7a. $\text{Ba}^{2+} + \text{SO}_4^{2-} \rightarrow \text{BaSO}_4$

7b. $\text{Pb}^{2+} + \text{CO}_3^{2-} \rightarrow \text{PbCO}_3$

8. net ionic eqn = $\text{Mg}^{2+} + 2\text{OH}^- \rightarrow \text{Mg(OH)}_2$

_____ 100

9a. $\text{Pb}^{2+} + \text{S}^{2-} \rightarrow \text{PbS}$

9b. $\text{Cu}^{2+} + \text{S}^{2-} \rightarrow \text{CuS}$

10. net ionic eqn = $\text{Ag}^+ + \text{Cl}^- \rightarrow \text{AgCl}$

_____ 100

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Rollin Morris Perkins



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Student's Solution Manual for Mathematical Investigations Phil DeMarois, Mercedes McGowan, Darlene Whitkanack, 2003-08-19

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Instructor's Solutions Manual [to Accompany] Mathematical Reasoning for Elementary Teachers, Third Edition Calvin Thomas Long, Duane W. DeTemple, 2003

Instructors' Manual for Criminal Investigation, Fifth Edition Wayne W. Bennett, 1998

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Handbook of EDP Auditing Michael Arthur Murphy, Michael A. Murphy, Xenia Ley Parker, 1990

Criminal Investigation Bruce Lawrence Berg, John J. Horgan, 1998-07

Study Guide/Selected Solutions Manual Julie R. Frentrup, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book.

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