

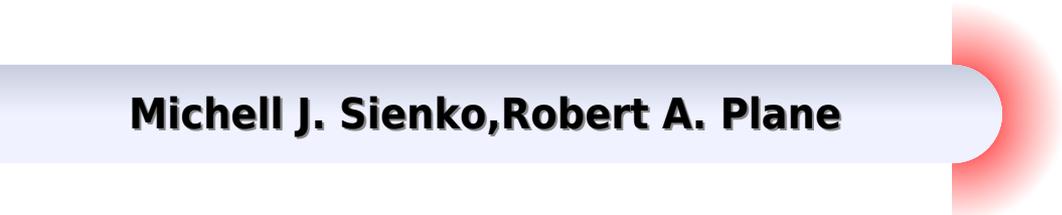
Mixed Stoichiometry Practice ([Answers](#))

1. How many moles of O_2 will be formed from 1.65 moles of KClO_3 ?
 2. How many moles of KClO_3 are needed to make 3.50 mol of KCl ?
 3. How many moles of KCl will be formed from 2.73 moles of KClO_3 ?
-
4. How much 0.20M H_3PO_4 is needed to react with 100mL of 0.10 M NaOH ?
-
-
-
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-
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-
-
-
5. If 36.0mL of H_3PO_4 reacts exactly with 80.0 mL of 0.500M NaOH , what is the concentration of the phosphoric acid?

15.0g of calcium hydroxide was added to 75.0mL of 0.500M HCl . How many grams of calcium chloride is formed? How much of the excess reagent was left over after the reaction has completed?

Mixed Stoichiometry Practice Answers

Michell J. Sienko, Robert A. Plane



Mixed Stoichiometry Practice Answers:

High School Chemistry Unlocked The Princeton Review, 2016-10-18 UNLOCK THE SECRETS OF CHEMISTRY with THE PRINCETON REVIEW High School Chemistry Unlocked focuses on giving you a wide range of key lessons to help increase your understanding of chemistry With this book you ll move from foundational concepts to complicated real world applications building confidence as your skills improve End of chapter drills will help test your comprehension of each facet of chemistry from atoms to alpha radiation Don t feel locked out Everything You Need to Know About Chemistry Complex concepts explained in straightforward ways Walk throughs of sample problems for all topics Clear goals and self assessments to help you pinpoint areas for further review Guided examples of how to solve problems for common subjects Practice Your Way to Excellence 165 hands on practice questions seeded throughout the chapters and online Complete answer explanations to boost understanding Bonus online questions similar to those you ll find on the AP Chemistry Exam and the SAT Chemistry Subject Test High School Chemistry Unlocked covers Building blocks of matter Physical behavior of matter Chemical bonding Chemical reactions Stoichiometry Solutions Acids and bases Equilibrium Organic chemistry Radioactivity and more *Chemistry* Greg Curran, 2011 Covers all the topics in a typical one year high school chemistry curriculum

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