

Mixed Gas Law Worksheet

Answers available at end

Name: _____ (show your work) period: _____

Convert:

100 mmHg = _____ kPa

300 mmHg = _____ atm

120 atm = _____ mmHg

300 torr = _____ kPa

25 kPa = _____ mmHg

25 mmHg = _____ atm

300°C = _____ K

150K = _____ °C

5000°C = _____ K

300K = _____ °C

-255°C = _____ K

3500K = _____ °C

200°C = _____ °F

95°F = _____ °C = _____ K

45°F = _____ °C = _____ K

- 1) A 70. liter sample of gas initially at 280°C is allowed to cool at constant pressure, what will the new volume be at 0°C?

$P_1 =$ _____ $P_2 =$ _____

$V_1 =$ _____ $V_2 =$ _____

$T_1 =$ _____ $T_2 =$ _____

- 2) A 4.0 liter sample of gas initially at 25°C is allowed to cool at constant pressure, what will the new volume be at -5°C?

$P_1 =$ _____ $P_2 =$ _____

$V_1 =$ _____ $V_2 =$ _____

$T_1 =$ _____ $T_2 =$ _____

- 3) A buoyancy vest (used in scuba diving) is filled to a volume of 2.3 liters at a pressure of 1 atmosphere (at the surface where the temperature is a balmy 300K), what will the volume of air be in the vest when the diver dives to a depth of about 90 feet where the temperature is 17°C and the pressure is 4 atmospheres?

$P_1 =$ _____ $P_2 =$ _____

Mixed Gas Law Worksheet Answers

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