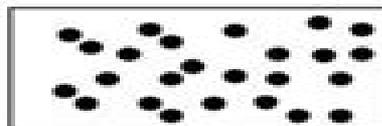


## Elements, Compounds, and Mixtures

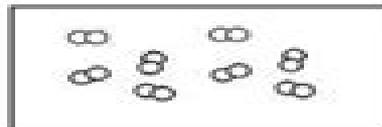
### Introduction

**Elements** are the simplest and most basic forms of matter. Elements cannot be broken down into any simpler substance – all known matter in the universe is made of elements.

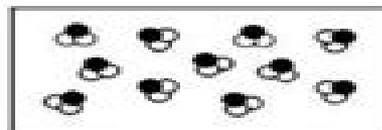
Some elements are made up of single, individual atoms.



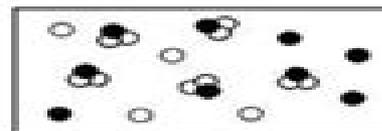
Other elements are made of **molecules**, which have two or more atoms joined together. In each of these examples, shown to the right, the elements are composed of only *one type of atom*.



When atoms of two or more different elements chemically combine, they form a new substance with completely different properties. These are known as **compounds**. Similar to elements, all of the molecules in a compound are identical to each other. Both elements and compounds are **pure substances**.

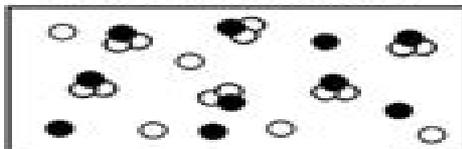


Elements and compounds do not always chemically react with each other. If different elements or compounds are placed together, but do not react, they are considered a **mixture**. Elements and compounds in mixtures retain their individual properties.

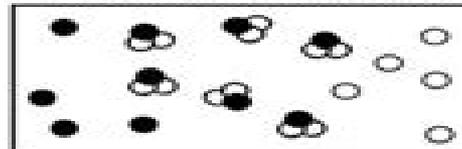


**Homogeneous mixtures** are uniform – they have the same composition throughout. **Heterogeneous mixtures** are not uniform. In a heterogeneous mixture, some areas may have higher or lower concentrations of the individual substances.

**Homogeneous Mixture**



**Heterogeneous Mixture**



# Mixtures Chemistry Practice Tests

**Robert S. Boikess**



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