

Unit 6 – Worksheet 3

Ionic Compounds

Properties

Basic structural unit

1. Give the name of the following simple binary ionic compounds.

- Na_2O
- K_2S
- MgCl_2
- CaBr_2
- BaI_2
- Al_2S_3
- CaBr
- AgF

2. Give the name of the following simple binary ionic compounds.

- Na_3N
- K_2O
- AgBr
- MgI_2
- SrO

3. Write the formula for the following binary ionic compounds.

- lithium bromide
- sodium iodide
- silver sulfide
- cesium oxide
- beryllium iodide
- barium hydride
- aluminum fluoride
- potassium oxide

4. Write the formula for these ionic substances.

- silver oxide
- aluminum sulfide
- sodium nitride
- barium chloride
- strontium hydride

Modeling Chemistry Unit 6 Test

Thomas Griffiths



Modeling Chemistry Unit 6 Test:

Adopting the Beat of Phrase: An Mental Symphony within **Modeling Chemistry Unit 6 Test**

In a global eaten by screens and the ceaseless chatter of quick conversation, the melodic splendor and psychological symphony created by the published term often disappear into the back ground, eclipsed by the persistent noise and disturbances that permeate our lives. However, set within the pages of **Modeling Chemistry Unit 6 Test** a charming fictional prize overflowing with organic thoughts, lies an immersive symphony waiting to be embraced. Constructed by a wonderful composer of language, that interesting masterpiece conducts readers on a psychological trip, skillfully unraveling the concealed tunes and profound influence resonating within each cautiously constructed phrase. Within the depths with this moving assessment, we shall investigate the book is main harmonies, analyze its enthralling writing style, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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