

Acids and bases end of unit worksheet

1. Complete the table below. I can identify properties of acids and bases.

	Acids	Bases
Taste	Sour	bitter
pH range	0-7	7-14
Color with litmus paper	pink/red	blue
Are they electrolytes? (yes or no)	yes	yes
What do they react with?	carbonates metals bases	fats+oils acids
What ion do they produce?	H^+ or H_3O^+	OH^-
How do they feel?	corrosive (itch, burn)	Slippery
Color with phenolphthalein	colorless	pink

2. I can list uses of acids in industry. List two uses of acids in industry.

Cleaning concrete (HCl) Making paint, plastics, fertilizers
producing batteries

3. I can list examples of common acids and bases found in a home or store.

List 2 common acids and 2 common bases you might find in your home or at a store.

Acids: lemons, vinegar, sour candy, pop, aspirin, orange
juice, lemonade

Bases: drain cleaner, soap, baking soda, ammonia,
window cleaner, tums, antacids

4. I can name acids and bases. Name the following:

HF hydrofluoric acid HNO_2 Nitrous acid $Ca(OH)_2$ calcium hydroxide

Write the formula for the following: (Check your charges!)

Magnesium hydroxide $Mg(OH)_2$ Hydrobromic acid HBr carbonic acid H_2CO_3

5. I can define and identify Arrhenius acids and bases. Label the following as an acid (A), base (B) or neutral (N).

H_3PO_4 A HCl A $BaBr_2$ N $LiOH$ B

6. I can label an acid as mono, di, or tri protic. Label the following as mono, di, or triprotic.

H_2SO_4 diprotic H_3PO_4 triprotic HNO_3 monoprotic HBr monoprotic

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2.1 ,1998 Improving High School Students' Performance in Chemistry with a Hands-on Approach Mary L. Fredell,1998

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