

## Released AP Multiple Choice Questions – Acids &amp; Bases / Aqueous Equilibrium

|   |   |
|---|---|
| D | 33. The pH of 0.1-molar ammonia is approximately<br>(A) 1<br>(B) 4<br>(C) 7<br>(D) 11<br>(E) 14   |
| E | 48. Which of the following ions is the strongest Lewis acid?<br>(A) $\text{Na}^+$<br>(B) $\text{Cl}^-$<br>(C) $\text{CH}_3\text{COO}^-$<br>(D) $\text{Mg}^{2+}$<br>(E) $\text{Al}^{3+}$   |
| C | 49. Each of the following can act as both a Brønsted acid and a Brønsted base EXCEPT<br>(A) $\text{HCO}_3^-$<br>(B) $\text{H}_2\text{PO}_4^-$<br>(C) $\text{NH}_4^+$<br>(D) $\text{H}_2\text{O}$<br>(E) $\text{HS}^-$   |
| B | 53. Which, if any, of the following species is in the greatest concentration in a 0.100-molar solution of $\text{H}_2\text{SO}_4$ in water?<br>(A) $\text{H}_2\text{SO}_4$ molecules<br>(B) $\text{H}_3\text{O}^+$ ions<br>(C) $\text{HSO}_4^-$ ions<br>(D) $\text{SO}_4^{2-}$ ions<br>(E) All species are in equilibrium and therefore have the same concentrations.   |
| E | 71. Which of the following reactions does NOT proceed significantly to the right in aqueous solutions?<br>(A) $\text{H}_3\text{O}^+ + \text{OH}^- \rightarrow 2 \text{H}_2\text{O}$<br>(B) $\text{HCN} + \text{OH}^- \rightarrow \text{H}_2\text{O} + \text{CN}^-$<br>(C) $\text{Cu}(\text{H}_2\text{O})_6^{2+} + 4 \text{NH}_3 \rightarrow \text{Cu}(\text{NH}_3)_4^{2+} + 4\text{H}_2\text{O}$<br>(D) $\text{H}_2\text{SO}_4 + \text{H}_2\text{O} \rightarrow \text{H}_3\text{O}^+ + \text{HSO}_4^-$<br>(E) $\text{H}_2\text{O} + \text{HSO}_4^- \rightarrow \text{H}_2\text{SO}_4 + \text{OH}^-$ |
| E | 75. If the acid dissociation constant, $K_a$ , for an acid HA is $8 \times 10^{-4}$ at $25^\circ\text{C}$ , what percent of the acid is dissociated in a 0.50-molar solution of HA at $25^\circ\text{C}$ ?<br>(A) 0.08%<br>(B) 0.2%<br>(C) 1%<br>(D) 2%<br>(E) 4%   |
| A | 34. All of the following species can function as Brønsted-Lowry bases in solution EXCEPT<br>(A) $\text{H}_2\text{O}$<br>(B) $\text{NH}_3$<br>(C) $\text{S}^{2-}$<br>(D) $\text{NH}_4^+$<br>(E) $\text{HCO}_3^-$   |

# General Equilibrium Ap Chemistry Multiple Choice

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