

POGIL: Lewis Dot Structures Part 1 (20 P)

How are electrons shared to create covalently bonded molecules?

Read This!

Why? Covalent bonds result from a *sharing* of electrons by two or more atoms (usually nonmetals). Lewis theory (Gilbert Newton Lewis, 1875-1946) focuses on the valence electrons, since the outermost electrons are the ones that are highest in energy and farthest from the nucleus, and are therefore the ones that are most exposed to other atoms when bonds form. Lewis dot diagrams for elements are a handy way of picturing valence electrons, and especially, which electrons are available to be shared in covalent bonds.

Bohr Model: Chlorine

7 valence electrons on outermost energy level



Lewis Dot diagram: Chlorine

7 valence electrons



chlorine
atoms



chlorine
molecule



Lewis Dot Diagrams Showing Valence Electrons for Individual Elements

1	2		3	4	5	6	7	8
H								He
Li	Be		B	C	N	O	F	Ne
Na	Mg	Transition metals	Al	Si	P	S	Cl	Ar

Atoms are grouped together by number of valence electrons; all atoms in Group 1 have 1 valence electron.

Lewis Structure Pogil

Richard Samuel Moog



Lewis Structure Pogil:

Organic Chemistry Suzanne M. Ruder, The POGIL Project, 2015-12-29 ORGANIC CHEMISTRY Chemists' Guide to Effective Teaching Norbert J. Pienta, Melanie M. Cooper, Thomas J. Greenbowe, 2005 For courses in Methods of Teaching Chemistry Useful for new professors chemical educators or students learning to teach chemistry Intended for anyone who teaches chemistry or is learning to teach it this book examines applications of learning theories presenting actual techniques and practices that respected professors have used to implement and achieve their goals Each chapter is written by a chemist who has expertise in the area and who has experience in applying those ideas in their classrooms This book is a part of the Prentice Hall Series in Educational Innovation for Chemistry **Exploring More Signature Pedagogies** Nancy L.

Chick, Aeron Haynie, Regan A. R. Gurung, 2023-07-03 What is distinctive about the ways specific disciplines are traditionally taught and what kinds of learning do they promote Do they inspire the habits of the discipline itself or do they inadvertently contradict or ignore those disciplines By analyzing assumptions about often unexamined teaching practices their history and relevance in contemporary learning contexts this book offers teachers a fresh way to both think about their impact on students and explore more effective ways to engage students in authentic habits and practices This companion volume to *Exploring Signature Pedagogies* covers disciplines not addressed in the earlier volume and further expands the scope of inquiry by interrogating the teaching methods in interdisciplinary fields and a number of professions critically returning to Lee S Shulman's origins of the concept of signature pedagogies This volume also differs from the first by including authors from across the United States as well as Ireland and Australia The first section examines the signature pedagogies in the humanities and fine arts fields of philosophy foreign language instruction communication art and design and arts entrepreneurship The second section describes signature pedagogies in the social and natural sciences political science economics and chemistry Section three highlights the interdisciplinary fields of Ignatian pedagogy women's studies and disability studies and the book concludes with four chapters on professional pedagogies nursing occupational therapy social work and teacher education that illustrate how these pedagogies change as the social context changes as their knowledge base expands or as online delivery of instruction increases *Process Oriented Guided Inquiry Learning (POGIL)* Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL's theoretical basis its implementations in diverse environments and evaluation of student outcomes

Modern NMR Spectroscopy in Education David Rovnyak, 2007 This book is intended to be a comprehensive resource for educators seeking to enhance NMR enabled instruction in chemistry This book describes a host of new modern laboratories and experiments **Extension of Lewis Structure Concepts to Open-shell and Excited- State Molecular Species** John Earl Carpenter, 1987 **Book Review Index - 2009 Cumulation** Dana Ferguson, 2009-08 Book Review Index provides quick access to reviews of books periodicals books on tape and electronic media representing a wide range of

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Molecules and Models Arne Haaland, 2008-03-06 This book provides a systematic description of the molecular structures and bonding in simple compounds of the main group elements with particular emphasis on bond distances bond energies and coordination geometries The description includes the structures of hydrogen halogen and methyl derivatives of the elements in each group some of these molecules are ionic some polar covalent The survey of molecules whose structures conform to well established trends is followed by representative examples of molecules that do not conform We also describe electron donor acceptor and hydrogen bonded complexes Chemists use models to systematize our knowledge to memorize information and to predict the structures of compounds that have not yet been studied The book provides a lucid discussion of a number of models such as the Lewis electron pair bond and the VSEPR models the spherical and polarizable ion models and molecular orbital calculations and it outlines the successes and failures of each **Lewis Structures and Molecular**

Geometry (computer File). Jeffrey R. Buell, 1968 **Writing Lewis Symbols and Lewis Structures** H. Anthony Neiding, James N. Spencer, 1994-01-01 **Valence And The Structure Of Atoms And Molecules** G.N. Lewis,

Introductory Chemistry Michael P. Garoutte, Ashley B. Mahoney, 2015-08-10 The ChemActivities found in Introductory Chemistry A Guided Inquiry use the classroom guided inquiry approach and provide an excellent accompaniment to any one semester Introductory text Designed to support Process Oriented Guided Inquiry Learning POGIL these materials provide a variety of ways to promote a student focused active classroom that range from cooperative learning to active student participation in a more traditional setting

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