

## Mole and Stoichiometry Mini-Lab

Calculations: Show **ALL** work (bridge/grid).

The **GIVEN** for calculations #4 and #5 should be measurements, not results from previous calculations.

4. How many moles of water are in the graduated cylinder?  
UNKNOWN: mol H<sub>2</sub>O      GIVEN: g H<sub>2</sub>O (data table)
5. How many molecules of water are in the graduated cylinder?  
UNKNOWN: molecules H<sub>2</sub>O  
GIVEN: g H<sub>2</sub>O (data table)
6. For an experiment, 4.75 moles of water are needed.
  - a) How many grams of water are needed?
  - b) How many milliliters of water would this be?  
UNKNOWN: g H<sub>2</sub>O      GIVEN: 4.75 mol H<sub>2</sub>O  
  
UNKNOWN: mL H<sub>2</sub>O      GIVEN: 4.75 mol H<sub>2</sub>O

## Mini Lab 1 apply Stoichiometry

**Sudheesh Valliyodan, Syed  
Wajih Ahmad Naqvi, Gireesh Kumar T.  
R., Annie Bourbonnais, Mark A.  
Altabet, Amal Jayakumar**

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**Recent Developments in Oxygen Minimum Zones Biogeochemistry** Sudheesh Valliyodan, Syed Wajih Ahmad Naqvi, Gireesh Kumar T. R., Annie Bourbonnais, Mark A. Altabet, Amal Jayakumar, 2023-12-18 **A Stoichiometry Unit** David Callaghan, 2004 **Stoichiometry Unit Project** Luann Marie Decker, 1998 *Stoichiometry* Sydney Young, 2012-01 Unlike some other reproductions of classic texts 1 We have not used OCR Optical Character Recognition as this leads to bad quality books with introduced typos 2 In books where there are images such as portraits maps sketches etc We have endeavoured to keep the quality of these images so they represent accurately the original artefact Although occasionally there may be certain imperfections with these old texts we feel they deserve to be made available for future generations to enjoy

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