

# MSCE CHEMISTRY PRACTICAL ANSWERS

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## Question 3

Table 1: Temperature Measurements

Measurement	Temperature (°C)
Initial temperature	25°C
Final temperature	45°C
Temperature change	20°C

Type of reaction: Exothermic

Reason: The temperature increased, indicating heat was released during the reaction.

### Energy Level Diagram

Reactants (higher energy) – Products (lower energy)

Energy released as heat.

Quantities to keep constant if repeated:

1. Volume of water (20 ml)
2. Number of sodium hydroxide pellets (5 pellets)

Possible source of error:

Heat loss to the surroundings before reading the final temperature.

## Question 4

Table 2: Solubility and Litmus Paper Results

Compound	Solubility in Water	Colour on Blue Litmus Paper
A	Soluble	No change (blue)
B	Soluble	Turns red
C	Insoluble	No change (blue)

Identifying the compounds:

- A: Ethanol
- B: Ethanoic acid
- C: Hexane

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**Padhraic Smyth**



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