

- 1 The table shows the melting points, boiling points and electrical conductivities of six substances **D**, **E**, **F**, **G**, **H** and **I**.

substance	melting point / °C	boiling point / °C	electrical conductivity when solid	electrical conductivity when liquid
D	1610	2230	non-conductor	non-conductor
E	801	1413	non-conductor	good conductor
F	-119	43	non-conductor	non-conductor
G	1535	2750	good conductor	good conductor
H	114	184	non-conductor	non-conductor
I	-210	-196	non-conductor	non-conductor

Choose substances from the table which match the following descriptions. Each substance may be used once, more than once or not at all.

- (a) Which substance is a liquid at 25 °C?

..... [1]

- (b) Which substance is a gas at 25 °C?

..... [1]

- (c) Which **three** substances contain simple molecules?

..... [3]

- (d) Which substance could be a metal? Give a reason for your answer.

substance

reason

..... [2]

- (e) Which substance has a macromolecular structure? Give **two** reasons for your answer.

substance

reason 1

reason 2

[3]

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