

## CALCULATING THE DIFFERENT ISOTOPES

### PURPOSE

The purpose of this experiment was to become familiar with isotopes and understand important information about them. This includes things like how different isotopes vary in mass and learning how to calculate corresponding relative mass and experimental mass.

### HYPOTHESIS

If asked to give the mass of a random piece of candy from you bag, what do you think the mass of the selected candy would be?

### PROCEDURE

1. Wait until the instructor hands you four bags of M&M's, there should be 2 brown bags and 2 yellow ones.
2. Before opening the bags, grab a plate in order to keep all the pieces of candy from scattering and to help with organization.
3. After pouring out all the pieces of candy, separate the M&M's by color and size.
4. After each color and size is in its own pile count the number in each pile and then record your data.
5. Next, multiply the number of small candies by its corresponding relative mass, located at the top of the instruction page, and do this for each color.
6. Repeat the step above except use the large M&M's and be sure to use to corresponding relative mass of the large candy.
7. For each color add the product received for the large candies and the product received for the small candies.
8. Then divide the sum calculated in step 7 by the total amount of small and large candies. (This will be you experimental mass)
9. After repeating this for all the colors, be sure to collect data from all the rest of the class and place it onto the chart provided.

### DATA

#### Green M&M

9 small pieces and 0 large pieces

Physical properties include the being green, solid, and a hard outer shell.

#### Blue M&M

4 small pieces and 4 large pieces

Physical properties include being blue and a solid.

#### Orange M&M

10 small pieces and 2 large pieces

# Introduction To Candy Lab Report

**Annelies Wilder-Smith**



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