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# > Coursebook answers

## Chapter 16

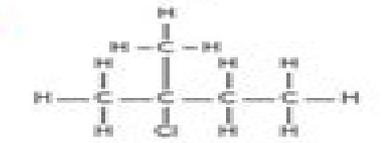
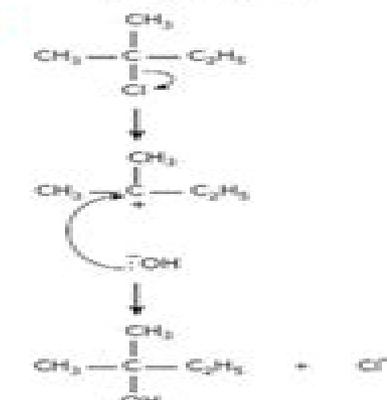
### Science in context

Discuss as a group how the information about the use of halogenoalkanes demonstrates how some scientific developments can have both good and bad effects on society and the environment.

- Having read the Science in Context passage, ask learners to shut their Coursebook. Then conduct a quick activity asking individual learners to give a specific example of a beneficial use of halogenoalkanes or a problematic use, until the class runs out of suggestions.
- You might then ask for a class vote, show of hands, on whether the halogenoalkanes have benefitted or hindered human progress in the last century.

### Self-assessment questions

- 1
- propene and chlorine
  - $\text{OH}^-$  ions are negatively charged so are attracted more strongly than neutral water molecules to the partially positively charged carbon atoms in halogenoalkanes.
  - The hydrolysis of halogenoalkanes produces halide ions; the rate of their formation can be monitored by using silver nitrate solution. The silver halide precipitates make the reaction mixture cloudy. The ionic equations for the formation of the precipitates are:  
 $\text{Ag}^+(\text{aq}) + \text{Cl}^-(\text{aq}) \rightarrow \text{AgCl}(\text{s})$   
 $\text{Ag}^+(\text{aq}) + \text{Br}^-(\text{aq}) \rightarrow \text{AgBr}(\text{s})$   
 $\text{Ag}^+(\text{aq}) + \text{I}^-(\text{aq}) \rightarrow \text{AgI}(\text{s})$
  - Both ammonia and amines contain a nitrogen atom with a lone pair of electrons that is available to donate.
  - triethylamine ( $(\text{CH}_3\text{CH}_2\text{CH}_2)_3\text{N}$ )

- 2
- 
  - 
  - 
- 3
- $\text{CH}_3\text{CHBrCH}_3 + \text{NaOH}(\text{ethanol}) \rightarrow \text{CH}_2 = \text{CHCH}_3 + \text{H}_2\text{O} + \text{NaBr}$
  - propene
  - $\text{BrCH}_2\text{CH}_2\text{CH}_3 + \text{NaOH}(\text{ethanol}) \rightarrow \text{CH}_2 = \text{CHCH}_3 + \text{H}_2\text{O} + \text{NaBr}$   
(1-bromopropane)

# Modern Chemistry Text Answers Chapter 16

**Thirumalaisamy P.  
Velavancorresponding**



## **Modern Chemistry Text Answers Chapter 16:**

Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 *Chemical Principles Student's Study Guide & Solutions Manual* John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better The solutions manual section includes detailed answers and explanations to the odd numbered exercises in the text Study and Problem Solving Guide to Accompany Principles of Modern Chemistry, Oxtoby/Nachtrieb Wade A. Freeman, 1986 **Introduction to Modern Chemistry** Edward Florian Neuzil, 1968 *Chemical Principles Study Guide/Solutions Manual* John Krenos, Joseph Potenza, Carl Hoeger, 2007-01-18 Written for general chemistry courses *Chemical Principles* helps students develop chemical insight by showing the connection between chemical principles and their applications **Molecular Driving Forces** Ken Dill, Sarina Bromberg, 2010-10-21 *Molecular Driving Forces* Second Edition E book is an introductory statistical thermodynamics text that describes the principles and forces that drive chemical and biological processes It demonstrates how the complex behaviors of molecules can result from a few simple physical processes and how simple models provide surprisingly accurate insights into the workings of the molecular world Widely adopted in its First Edition *Molecular Driving Forces* is regarded by teachers and students as an accessible textbook that illuminates underlying principles and concepts The Second Edition includes two brand new chapters 1 Microscopic Dynamics introduces single molecule experiments and 2 Molecular Machines considers how nanoscale machines and engines work The Logic of Thermodynamics has been expanded to its own chapter and now covers heat work processes pathways and cycles New practical applications examples and end of chapter questions are integrated throughout the revised and updated text exploring topics in biology environmental and energy science and nanotechnology Written in a clear and reader friendly style the book provides an excellent introduction to the subject for novices while remaining a valuable resource for experts *Chemistry* Leonard W. Fine, 1972

**CHEMISTRY SILBERBERG**, 2003 *Materials Science for Engineers* J.C. Anderson, Keith D. Leaver, Rees D. Rawlings, Patrick S. Leever, 2004-06-01 This fifth edition of a successful textbook continues to provide students with an introduction to the basic principles of materials science over a broad range of topics The authors have revised and updated this edition to include many new applications and recently developed materials The book is presented in three parts The first section discusses the physics chemistry and internal structure of materials The second part examines the mechanical properties of materials and their application in engineering situations The final section presents the electromagnetic properties of materials and their application Each chapter begins with an outline of the relevance of its topics and ends with problems that require an understanding of the theory and some reasoning ability to resolve These are followed by self assessment questions which test students understanding of the principles of materials science and are designed to quickly cover the subject area of the chapter This edition of *Materials Science for Engineers* includes an expanded treatment of many

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**The Educational Times, and Journal of the College of Preceptors** ,1893 *Modern Textbook of Organic Chemistry* Gwynn Pennant Ellis,1966 **A Course in Colloid Chemistry** D. A. Fridrikhsberg,1986 **General College Chemistry** Charles William Keenan,Donald C. Kleinfelter,Jesse Hermon Wood,1980 *Chemistry* John S. Phillips,Cheryl Wistrom,2000 *Nature* Sir Norman Lockyer,1911 **General Chemistry** James E. Brady,Gerard E. Humiston,1986 The Fifth Edition retains the pedagogical strengths that made the previous editions so popular and has been updated reorganized and streamlined Changes include more accessible introductory chapters with greater stress on the logic of the periodic table earlier introduction of redox reactions greater emphasis on the concept of energy a new section on Lewis structures earlier introduction of the ideal gas law and a new development of thermodynamics Each chapter ends with review questions and problems

**Chemical News and Journal of Physical Science** ,1913

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web circle the correct answers the mass of the system a b is m 2m the force that accelerates a b is the weight of a b the weight of b is mg 2 acceleration of a b is less than mg 2mg g g more than g use a to show the acceleration of a b as a fraction of conceptual physics

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web t t toward center of circle yes yes yes f f because centripetal acceleration is not zero n n yes provides centripetal force for circular motion conceptual physics

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web dec 11 2012 1 kg consider the acceleration of the cart when the applied force is due to a 10 n iron weight attached to a string draped over pulley will the cart accelerate as before at 10 m/s<sup>2</sup> the answer is no because the mass being accelerated is the mass of the cart plus the mass of the piece of iron that pulls it both masses accelerate

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web dec 12 2012 1 draw sample force of shown hand 3 a which the b which the 4 it is important to distinguish between

force and velocity vectors force vectors combine with other force vectors and velocity vectors combine with other velocity vectors do velocity vectors combine with force vectors no 5

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web circle the correct answers 5 we see that tension in a rope is dependent on independent of the length of the rope so the length of a vector representing rope tension is dependent on independent of the length of the rope concept development 2 2 practice page

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web mar 4 2013 the distance between the balls decreases the wavelength decreases just as the distance between the balls in question 5 decreases 30 m 30 cm 1 m s

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web our intuitive definition of force that is a push or a pull is a good place to start we know that a push or pull has both magnitude and direction therefore it is a vector quantity and can vary considerably in each regard for example a cannon exerts a strong force on a cannonball that is launched into the air

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