

Molar Mass

Worksheet-Answer Key

Name _____ Date _____

Exercise 1. How much does one mole weigh (molar mass/molecular weight) for the following? Show work: which numbers are added/multiplied, how many decimal places are used from the periodic table, and your result.

Na ___ 23g ___

22.99 → 23 g

NaF ___ 42g ___

23 + 19 = 42 g

Zn ___ 65.4 g ___

65.38 → 65.4 g

CO ___ 28 g ___

12 + 16 = 28 g

solid iron ___ 55.9 g ___

Fe 55.85 → 55.9 g

CO₂ ___ 44 g ___

12 + 2×16 = 12 + 32 = 44 g

Ca²⁺ (aq) ___ 40 g ___

same as Ca 40.08 → 40 g

CH₄ ___ 16 g ___

12 + 4×1 = 12 + 4 = 16

CaCO₃ ___ 100 g ___

40 + 12 + 3×16 = 100g

C₄H₁₀ ___ 58 g ___

4×12 + 10×1 = 58g

Molar Mass Practice Answer Sheet

N Noddings



Molar Mass Practice Answer Sheet:

Molar Mass Practice Answer Sheet Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Molar Mass Practice Answer Sheet**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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