

Molarity Practice Worksheet

Find the molarity of the following solutions:

- 1) 0.5 moles of sodium chloride is dissolved to make 0.05 liters of solution.

- 2) 0.5 grams of sodium chloride is dissolved to make 0.05 liters of solution.

- 3) 0.5 grams of sodium chloride is dissolved to make 0.05 mL of solution.

- 4) 734 grams of lithium sulfate are dissolved to make 2500 mL of solution.

- 5) 6.7×10^{-2} grams of $\text{Pb}(\text{C}_2\text{H}_3\text{O}_2)_4$ are dissolved to make 3.5 mL of solution.

- 6) I have two solutions. In the first solution, 1.0 moles of sodium chloride is dissolved to make 1.0 liters of solution. In the second one, 1.0 moles of sodium chloride is added to 1.0 liters of water. Is the molarity of each solution the same? Explain your answer.

Molarity Practice Unit Lesson Key

M Lipman



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Enjoying the Beat of Phrase: An Psychological Symphony within **Molarity Practice Unit Lesson Key**

In a world used by displays and the ceaseless chatter of quick communication, the melodic beauty and psychological symphony created by the written word usually fade in to the background, eclipsed by the constant noise and interruptions that permeate our lives. However, nestled within the pages of **Molarity Practice Unit Lesson Key** a wonderful fictional treasure filled with natural emotions, lies an immersive symphony waiting to be embraced. Constructed by an outstanding composer of language, this interesting masterpiece conducts visitors on a mental trip, well unraveling the hidden tunes and profound influence resonating within each cautiously crafted phrase. Within the depths of the moving evaluation, we can explore the book is main harmonies, analyze its enthralling publishing design, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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