The Mole Webquest

1. Use the following website to answer the set of questions:
[illegible]

   1). A mole of anything is how many of that thing? (plus the notation)
   6.02 × 10^{23}

   2). Why do Urban different amounts of things can add equal one mole? (think about the weight of a mole's elephants vs a dozen eggs)

A mole is an amount of something. Different things can have different molecular masses but still be equal to one mole.

   3). Why do we want to use the concept of moles?

It is necessary to know how many molecules of a sample substance you have.

   4). Crossing (specific number of moles) are then converted to the number of molecules or masses and vice versa.

   5). How many grams of water are in one mole of water?

18 grams.

   6). How many molecules of water are in one mole of water?

6.02 × 10^{23} water molecules.

2. Use the following website to answer the next set of questions:
[illegible]

   7). What is the molar mass (mass of one mole) of CO_2?

   44.01 grams

Mole Webquest Answer Key

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Mole Webquest Answer Key:

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