

## Study Guide – Unit 1 Periodic Table ANSWERS

### Structure of an atom

- List the three subatomic particles which make up an atom. Include their location, charge, and atomic mass.
  - Electron Negative charge Electron Cloud
  - Proton Positive charge Nucleus
  - Neutron Neutral charge Nucleus
- In a neutral atom, which two subatomic particles are equal in number to each other?  
Protons and Electrons
- What does the atomic number of an element indicate?  
The number of protons in the atom of that element
- What is an isotope?  
Atoms with the same number of protons, but different masses due to a different number of neutrons
- Will two different atoms of carbon always have the same number of protons?  
Yes

### Element key

13
Al
26.982
Aluminum

- What are the number of protons, neutrons, and electrons in an atom of aluminum?
  - Protons = 13
  - Neutrons = 14
  - Electrons = 13

### Bohr Model

- How do you know how many energy levels (orbits) a Bohr model will have?
  - By the period the element is located in
- How many electrons can each energy level hold?  
 $1^{\text{st}} - 2$                        $2^{\text{nd}} - 8$                        $3^{\text{rd}} - 18$

### The Periodic Table

- How is the periodic table arranged? Increasing atomic number
- What are the vertical columns  $\updownarrow$  called? Groups or Families
- What are the horizontal rows  $\leftrightarrow$  called? Periods
- Why are elements put into groups? Same number of valence electrons
- What is a valence electron? Number of electrons on the outer energy level (shell)
- Know how many valence electrons each group has
- Which group of metals are highly reactive and why? Alkali Metals. Only 1 ve<sup>-</sup> to lose
- Which group of non-metals are highly reactive and why? Halogens. Only 1 ve<sup>-</sup> to gain
- Name some physical properties of metals: luster, ductile, malleable, conductor of heat and electricity
- Name some physical properties of non metals: no luster, brittle, not a good conductor
- What are metalloids and where are they found on the periodic table? Share properties of metals and metalloids. On zig-zag line starting at Boron,

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## **Table of Contents Excel Lesson 3 Study Guide With Answers**

1. Understanding the eBook Excel Lesson 3 Study Guide With Answers
  - The Rise of Digital Reading Excel Lesson 3 Study Guide With Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Excel Lesson 3 Study Guide With Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Excel Lesson 3 Study Guide With Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Excel Lesson 3 Study Guide With Answers
  - Personalized Recommendations
  - Excel Lesson 3 Study Guide With Answers User Reviews and Ratings
  - Excel Lesson 3 Study Guide With Answers and Bestseller Lists

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  - Excel Lesson 3 Study Guide With Answers Enhanced eBook Features
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  - Interactive Elements Excel Lesson 3 Study Guide With Answers
8. Staying Engaged with Excel Lesson 3 Study Guide With Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Excel Lesson 3 Study Guide With Answers
9. Balancing eBooks and Physical Books Excel Lesson 3 Study Guide With Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Excel Lesson 3 Study Guide With Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Excel Lesson 3 Study Guide With Answers
  - Setting Reading Goals Excel Lesson 3 Study Guide With Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Excel Lesson 3 Study Guide With Answers
  - Fact-Checking eBook Content of Excel Lesson 3 Study Guide With Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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