

BSN EXAM I Study Guide

ACID-BASE IMBALANCE (12 QUESTIONS)

1. Explain the body's three regulators used to maintain a normal pH:
 - a. Buffers – act immediately!
 - i. Carbonic acid-bicarbonate
 - ii. HCl = strong acid
 - iii. H_2CO_3 = carbonic acid
 - iv. NaH_2CO_3 = sodium bicarb.
 - v. CO_2 = acid
 - vi. H^+ ions & H_2CO_3 (etc.) work to regulate pH
 - vii. Also, phosphates + proteins + hemoglobin buffers: bind & transport (ex: carry things to kidney to be excreted)
 1. *****If something goes in, something must go out = usually K^{+***}**
 - b. Respiratory System – works quickly (mins = hours)
 - i. PaCO_2 is only altered by respiratory causes
 1. = respirations = CO_2 elimination + = CO_2 in blood
 2. = respirations = CO_2 retention
 - ii. PaCO_2 is the best indicator of alveolar ventilation!
 - c. Renal System – slower: takes 2-3 days
 - i. HCO_3 is only altered by metabolic processes.
 - ii. Renal system usually conserves bicarb & excretes acid
 - iii. May be caused by gain/loss of acids OR gain/loss of HCO_3 /bicarbonate
1. List the normal value parameters for an arterial blood gas.
 - a. pH = 7.35 – 7.45
 - b. PaCO_2 = 35 – 45 mmHg partial pressure of carbon dioxide in blood
 - c. HCO_3 = 22 – 26 mEq/L
 - d. PaO_2 = 80 – 100 mmHg (on room air)
 - e. O_2 sat = $\geq 95\%$
2. Analyze the four different acid base imbalances (respiratory acidosis; respiratory alkalosis; metabolic acidosis; metabolic alkalosis) including for each:
 - a. Respiratory Acidosis: pH = / CO_2 =
 - i. Causes: = elimination of CO_2 from lungs
 1. Oversedation
 2. Anesthesia
 3. Head injuries
 4. Drug overdoses
 5. COPD
 6. Neuromuscular diseases (M. Gravis, Guillain-Barre)
 7. Mechanical ventilator: HYPOventilation
 8. Respiratory failure
 - ii. Clinical manifestations:
 1. CNS depression (change in LOC)

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