

- 5) If I have 7.7 moles of gas at a pressure of 0.09 atm and at a temperature of 56 °C, what is the volume of the container that the gas is in?

$$PV = nRT$$

$$V = \frac{nRT}{P} = \frac{7.7 \times 329 \times .0821}{.09 \text{ atm}} = 2310.9 \text{ L}$$

- 7) If I have 17 moles of gas at a temperature of 67 °C, and a volume of 88.89 liters, what is the pressure of the gas?

$$PV = nRT$$

$$P = \frac{nRT}{V}$$

$$\frac{17 \cdot .0821 \cdot 340}{88.89 \text{ L}} = 5.34 \text{ atm}$$

- 8) If I have an unknown quantity of gas at a pressure of 0.5 atm, a volume of 25 liters, and a temperature of 300 K, how many moles of gas do I have?

$$PV = nRT$$

$$n = \frac{PV}{RT}$$

$$\frac{.0821 \times 300}{.5 \text{ atm} \times 25 \text{ L}} = 1.97 \text{ mol}$$

- 9) If I have 21 moles of gas held at a pressure of 78 atm and a temperature of 900 K, what is the volume of the gas?

$$PV = nRT$$

$$V = \frac{nRT}{P}$$

$$\frac{21 \cdot 0.0821 \cdot 900}{78} = 19.89 \text{ L}$$

- 10) If I have 1.9 moles of gas held at a pressure of 5 atm and in a container with a volume of 50 liters, what is the temperature of the gas?

$$PV = nRT$$

$$T = \frac{PV}{nR}$$

$$\frac{5 \text{ atm} \cdot 50 \text{ L}}{1.9 \text{ mol} \cdot .0821} = 1602.67 \text{ K}$$

- 11) If I have 2.4 moles of gas held at a temperature of 97 °C and in a container with a volume of 45 liters, what is the pressure of the gas?

$$PV = nRT$$

$$P = \frac{nRT}{V}$$

$$\frac{2.4 \times .0821 \cdot 370}{45} =$$

1.62 atm

Gas Law Problems Answers

A Gutmann



Gas Law Problems Answers:

Gas Law Problems Answers Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Gas Law Problems Answers**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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