

Gas Laws Worksheet - CP

SHOW YOUR WORK for full credit!!!

Use the Combined Gas Law (or its component formulas) to answer the following questions:

21) For questions (a → f) circle the answer that describes what would happen in each scenario:

- | | |
|--|---------------------------------|
| a) For Boyle's law if pressure increases, what happens to volume? | (Increase / Decrease / Neither) |
| b) For Boyle's Law, if volume decreases, what happens to pressure? | (Increase / Decrease / Neither) |
| c) For Charles's Law, if temperature increases, what happens to volume? | (Increase / Decrease / Neither) |
| d) For Charles's Law, if volume decreases, what happens to temperature? | (Increase / Decrease / Neither) |
| e) For Gay-Lussac's Law, if pressure increases, what happens to temperature? | (Increase / Decrease / Neither) |
| f) For Gay-Lussac's Law, if temperature decreases, what happens to pressure? | (Increase / Decrease / Neither) |

22) If 22.5 L of nitrogen at 748 mm Hg are compressed to 725 mm Hg at constant temperature. What is the new volume?

23) Calculate the decrease in temperature when 6.00 L at 20.0 °C is compressed to 4.00 L.

24) The gases in a hair spray can are at a temperature of 27 °C and a pressure of 3 atm . If the gases in the can reach a pressure of 9 atm, the can will explode. To what temperature must the gases be raised in order for the can to explode? Assume constant volume.

25) A gas balloon has a volume of 106.0 liters when the temperature is 45.0 °C and the pressure is 740.0 mm of mercury. What will its volume be at 20.0 °C and 780 .0 mm of mercury pressure?

26) What pressure is required to compress 196.0 liters of air at 1.00 atmosphere into a cylinder whose volume is 26.0 liters?

27) A gas occupies 900.0 mL at a temperature of 27.0 °C. What is the volume at 132.0 °C?

28) Maybelline Cousteau's backup oxygen tank reads 900 mmHg while on her boat, where the temperature is 27 °C. When she dives down to the bottom of an unexplored methane lake on a recently-discovered moon of Neptune, the temperature will drop down to -183 °C. What will the pressure in her backup tank be at that temperature?

29) If 10.0 liters of oxygen at STP are heated to 512 °C, what will be the new volume of gas if the pressure is also increased to 2.0 atm?

Gas Laws Worksheet And Answers

Melissa May Batchelor



Gas Laws Worksheet And Answers:

The book delves into Gas Laws Worksheet And Answers. Gas Laws Worksheet And Answers is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Gas Laws Worksheet And Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Gas Laws Worksheet And Answers
 - Chapter 2: Essential Elements of Gas Laws Worksheet And Answers
 - Chapter 3: Gas Laws Worksheet And Answers in Everyday Life
 - Chapter 4: Gas Laws Worksheet And Answers in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, this book will provide an overview of Gas Laws Worksheet And Answers. The first chapter will explore what Gas Laws Worksheet And Answers is, why Gas Laws Worksheet And Answers is vital, and how to effectively learn about Gas Laws Worksheet And Answers.
 3. In chapter 2, the author will delve into the foundational concepts of Gas Laws Worksheet And Answers. The second chapter will elucidate the essential principles that need to be understood to grasp Gas Laws Worksheet And Answers in its entirety.
 4. In chapter 3, this book will examine the practical applications of Gas Laws Worksheet And Answers in daily life. The third chapter will showcase real-world examples of how Gas Laws Worksheet And Answers can be effectively utilized in everyday scenarios.
 5. In chapter 4, this book will scrutinize the relevance of Gas Laws Worksheet And Answers in specific contexts. This chapter will explore how Gas Laws Worksheet And Answers is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, this book will draw a conclusion about Gas Laws Worksheet And Answers. This chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Gas Laws Worksheet And Answers.

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