

## Model 2 – Ternary Acids (Oxyacids)

Acid	Name of Acid in Aqueous Solution	Cation (+)	Polyatomic Anion (-)	Polyatomic Anion Name
$\text{HClO}_3$	Chloric acid	$\text{H}_3\text{O}^+$	$\text{ClO}_3^-$	Chlorate
$\text{H}_2\text{SO}_3$	Sulfurous acid	$2\text{H}_3\text{O}^+$	$\text{SO}_3^{2-}$	Sulfite
$\text{H}_2\text{SO}_4$	Sulfuric acid	$2\text{H}_3\text{O}^+$	$\text{SO}_4^{2-}$	Sulfate
$\text{H}_3\text{PO}_3$	Phosphorous acid	$\text{H}_3\text{O}^+$	$\text{PO}_3^{3-}$	Phosphite
$\text{H}_3\text{PO}_4$	Phosphoric acid	$3\text{H}_3\text{O}^+$	$\text{PO}_4^{3-}$	Phosphate
$\text{HNO}_3$	Nitric acid	$\text{H}_3\text{O}^+$	$\text{NO}_3^{-1}$	Nitrate
$\text{HNO}_2$	Nitrous acid	$\text{H}_3\text{O}^+$	$\text{NO}_2^{-1}$	Nitrite
$\text{H}_2\text{CO}_3$	Carbonic acid	$2\text{H}_3\text{O}^+$	$\text{CO}_3^{2-}$	Carbonate

6. Look at the formulas of the ternary acids in Model 2.

a. How are ternary acids different from binary acids in their structure?

While binary acids consist of two different elements, ternary acids have three different types of elements

b. What number do you think the prefix "ter-" refers to?

Three

7. When ternary acids are mixed with water, ions will form. Fill in the table above with the

ions that form. For example, when  $\text{HClO}_3$  is mixed with water, the  $\text{H}_3\text{O}^+$  and  $\text{ClO}_3^-$  ions form. Write the formulas of the ions that form when each of the acids in Model 2 is mixed with water.

Phosphorous acid:  $\text{H}_3\text{PO}_3$  + water yields  $\text{H}_3\text{O}^+$  and  $\text{PO}_3^{3-}$

Phosphoric acid:  $\text{H}_3\text{PO}_4$  + water yields  $3\text{H}_3\text{O}^+$  and  $\text{PO}_4^{3-}$

Nitrous acid:  $\text{HNO}_2$  + water yields  $\text{H}_3\text{O}^+$  and  $\text{NO}_2^{-1}$

8. Draw a Lewis structure for each of the polyatomic anions that form when each of the acids in Model 2 is mixed with water.

Phosphate:  $\text{PO}_4^{3-}$  (Lewis structure: central P atom bonded to 4 O atoms, one double bond, three single bonds with negative charges)

Sulfate:  $\text{SO}_4^{2-}$  (Lewis structure: central S atom bonded to 4 O atoms, two double bonds, two single bonds with negative charges)

Carbonate:  $\text{CO}_3^{2-}$  (Lewis structure: central C atom bonded to 3 O atoms, one double bond, two single bonds with negative charges)

9. Write the formula for each of the ions that form.

$\text{H}_3\text{O}^+$

10. Draw the Lewis structure for each of the polyatomic anions that form when each of the acids in Model 2 is mixed with water.

$\text{ClO}_3^-$

$\text{SO}_3^{2-}$

$\text{SO}_4^{2-}$

$\text{PO}_3^{3-}$

$\text{PO}_4^{3-}$

# Naming Acids Pogil Answer Key

**J Ma**

### **Naming Acids Pogil Answer Key:**

*The Etymology of Chemical Names*, 2019-10-08 Etymology of Chemical Names gives an overview of the development of the current chemical nomenclature tracing its sources and changing rules as chemistry progressed over the years This book is devoted to provide a coherent picture how the trivial and systematic names shall be used and how the current IUPAC rules help to reconcile the conflicting demands **Proton Chemistry** Robert A. Richardson, 1976 Alberta Authorized Resource for grade 12 ca 1980 1994 1947, □□□□□□□□

## **Naming Acids Pogil Answer Key** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Naming Acids Pogil Answer Key**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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web fsk ab 12 freigegeben 1941 nach dem ende des ersten abenteuers im sagenumwobenen land narnia leben die pevencie geschwister lucy georgie henley edmund skandard keynes peter william

*die chroniken von narnia 04 prinz kaspian von narnia 4 cds* - Jan 28 2022

web kaspian hofft auf das zauberhorn das ihm aus alten zeiten überlassen wurde um verstärkung zu rufen es kommen aber nicht die legendären königinnen und könige von narnia sondern vier kinder die gerade auf dem weg zur schule waren was können die schon ausrichten und was ist mit den königen aus alter zeit geschehen