

Nomenclature Skill Review: Naming Binary Compounds

Binary Ionic Compounds (Type I & II) & Review rules & for look at flowchart for guidance

Write the symbol for the ion:	Name the ion:
Lithium	Li^+
Strontium	Sr^{2+}
Bromide	Br^-
Cobalt (III)	Co^{3+}
Oxide	O^{2-}
Calcium	Ca^{2+}
Copper (II)	Cu^{2+}
Aluminum	Al^{3+}
Silver	Ag^+
Vanadium (V)	VO_4^{3-}
Write the formula:	Name the compound:
Cesium chloride	CsBr
Calcium fluoride	CaF_2
Rubidium sulfide	Rb_2S
Copper (II) oxide	CuO
Potassium iodide	KI
Titanium (IV) selenide	TiSe_2
Silver sulfide	Ag_2S
aluminum sulfate	$\text{Al}_2(\text{SO}_4)_3$
Gallium phosphate	$\text{Ga}_3(\text{PO}_4)_2$
Beryllium nitride	Be_3N_2

Binary Molecular Compounds (Type III) & Review rules & for look at flowchart for guidance

Write the formula:	Write the formula:	Name the compound:
2	Diboron hexafluoride	B_2F_6
2	Carbon monoxide	CO
3	Phosphorus pentabromide	PBr_5
4	Tritrogen heptafluoride	N_3F_7
5	Tetrasulfur decoxide	S_4O_{10}
6	Sulfur dioxide	SO_2
7		
8		
9		
10		

Naming Review Guide Answer Key

Lauren Gardner



Naming Review Guide Answer Key:

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The Balance Sheet, 1929

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Putting the Movement Back

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