

Activity #9

Answer Key

Key Questions:

1) Describe the difference between the distribution of electron domains and molecular geometry.

The distribution of electron domains considers all bonding and non-bonding domains. They will arrange so that all domains are as far apart as possible. The molecular geometry considers only bonding domains.

2) List the steps necessary to determine the bond angles in a molecule.

- Determine the Lewis structure
- Arrange e^- domains into the correct geometry for that number of domains
- Check for influence of lone pairs

3) How can the geometry of a molecule be determined from the distribution of electron domains?

By looking at the arrangement of only the bonding domains

4) Label the bond angles in the SF₄ molecule shown in Table 1.



5) How many different kinds of molecular geometries are possible for a molecule with a tetrahedral distribution of electron domains?

There are four molecular geometries:

Linear	(3 NB, 1 B)
Bent	(2 NB, 2 B)
Trigonal pyramidal	(1 NB, 3 B)
Tetrahedral	(0 NB, 4 B)

*** If 4 non-bonding domains, like Cl⁻, it wouldn't be a molecule, so no molec. geometry ***

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M Tight



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