

## Mixed Stoichiometry Practice

Name \_\_\_\_\_

Date \_\_\_\_\_ Period \_\_\_\_\_

Write and/or balance the following equations (remember the diatomic elements and to cross-check charges for ionic compounds!!!) Use the mole ratios from the balanced equations to solve the following stoichiometry problems. Use units and labels in all conversions, and round your answer to sig figs.

1. Potassium chlorate decomposes into potassium chloride and oxygen gas.

**Balanced Equation:**

2. How many moles of oxygen are produced when 3.0 moles of potassium chlorate decompose completely?

3. Butane ( $C_4H_{10}$ ) undergoes combustion.

**Balanced Equation:** \_\_\_\_\_  $C_4H_{10}(l)$  + \_\_\_\_\_  $O_2(g)$  → \_\_\_\_\_  $CO_2(g)$  + \_\_\_\_\_  $H_2O(l)$

4. How many grams of  $CO_2$  are produced when 60 g of  $O_2$  are reacted with an excess of butane?

5. Water decomposes into hydrogen gas and oxygen gas by electrolysis.

**Balanced Equation:**

6. How many grams of hydrogen will be produced when 6.0 moles of oxygen are produced?

7. How many grams of water are required to produce 9.00 grams of hydrogen?

# Mixed Stoichiometry Problems Answer Key

**RJ Alexander**



**Mixed Stoichiometry Problems Answer Key:**

## Decoding **Mixed Stoichiometry Problems Answer Key**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Mixed Stoichiometry Problems Answer Key**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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