

ATOMIC MODELS

Atomic Models

Reading Passage

Read the passage carefully before answering the questions that follow.

The development of atomic models has been a central pursuit in the history of science. Starting from an understanding of the fundamental building blocks of matter, the earliest and simplest atomic models, such as Democritus' indivisible model, depicted atoms as indivisible spheres, primarily based on limited experimental data. These models focused on the concept of atoms as discrete particles or particles of matter, but they did not address the internal structure of atoms.

As scientific techniques advanced in the 19th century, scientists like J.J. Thomson proposed the plum pudding model, where electrons were embedded within a positively charged sphere. Later, Ernest Rutherford's gold foil experiments revealed that atoms contained mostly of empty space with a dense, positively charged nucleus in the center. This led to the planetary model, which depicted electrons orbiting the nucleus like planets around a sun.

However, the planetary model could not explain many observations, such as the spectral lines of elements. This challenge led to the development of quantum mechanical models in the early 20th century. These models describe electrons not as particles orbiting in fixed paths but as probability "clouds." Modern atomic models use the Schrödinger equation to describe electron behavior. These models recognize that electrons do not have fixed paths or discrete orbits. Instead, they exist in regions of probability, called orbitals, which represent the likelihood of finding an electron in a particular region.

Experimental evidence like spectroscopy, which studies the interaction of matter with electromagnetic radiation, and scattering experiments, which probe atomic structure, have been crucial in refining atomic models. The evolution of these models illustrates the scientific process of understanding matter and its components through continuous observation and experimentation. The continuous refinement of atomic models reflects the scientific progress driven by experimental evidence, helping our comprehension of the intricate fundamental structure.

True or False: Atomic Models and Scientific Evidence

Circle whether each statement is True or False.

1. Democritus' indivisible model portrayed atoms as indivisible spheres.

2. True

3. False

2. The plum pudding model suggested that electrons were embedded within positively charged spheres.

3. True

3. False



Making Atomic Models Answer Key

**Eddy Arnold, Daniel M. Himmel, Michael
G. Rossmann**

Making Atomic Models Answer Key:

International Tables for Crystallography, Volume F Eddy Arnold, Daniel M. Himmel, Michael G.

Rossmann, 2012-03-05 International Tables for Crystallography Volume F is an expert guide to macromolecular crystallography for the structural biologist. It was commissioned by the International Union of Crystallography in recognition of the extraordinary contributions that knowledge of macromolecular structure has made and will make to the analysis of biological systems from enzyme catalysis to the workings of a whole cell. The volume covers all stages of a crystallographic analysis from the preparation of recombinant proteins through crystallization, diffraction data collection, phase determination, structure validation and structure analysis. Although the volume is written for experienced scientists, it is recognized that the reader is more likely to be a biologist interested in structure than a classical crystallographer interested in biology. Thus there are chapters on the fundamentals, history and current perspectives of macromolecular crystallography as well as on useful programs and databases such as the Protein Data Bank. Each chapter is written by one or more internationally recognized experts. This second edition features 19 new articles and many articles from the first edition have been revised. The new articles cover topics such as standard definitions for quality indicators, expression of membrane proteins, protein engineering, high throughput crystallography, radiation damage, merohedral twinning, low resolution ab initio phasing, robotic crystal loading, whole cell X ray diffraction imaging and halogen interactions in biological crystal structures. There are also new articles on relevant software including software for electron microscopy. These enhancements will ensure that Volume F continues to be a key reference for macromolecular crystallographers and structural biologists. More information on the series can be found at <http://it.iucr.org>

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Introduces new chemistry concepts and provides activities so that students can practice and grasp the concepts Key terms are highlighted in the text as well as in a comprehensive glossary Answer keys are included

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