

Name Mel

Locker # \_\_\_\_\_

**Bonding & Molecular Structure****LEWIS STRUCTURES**

Indicate the total # of VALENCE electrons for each molecule (VE's). Calculate the number of bonded electrons from the skeleton (BE's). Calculate the Remaining Electrons (RE's = VE's - BE's) and distribute the remaining electrons onto the skeletal structure. Finally, satisfy the octets by creating double and triple bonds. Your final structures should be given below and your scratch paper should clearly show your work.

$\text{F}_2$ VE's = <u>14</u> BE's = <u>2</u> RE's = <u>12</u> $\text{:}\ddot{\text{F}}-\ddot{\text{F}}\text{:}$	$\text{H}_2$ VE's = <u>2</u> BE's = <u>2</u> RE's = <u>0</u> $\text{H}-\text{H}$	$\text{HF}$ VE's = <u>8</u> BE's = <u>2</u> RE's = <u>6</u> $\text{H}-\ddot{\text{F}}\text{:}$	$\text{NH}_3$ VE's = <u>8</u> BE's = <u>6</u> RE's = <u>2</u> $\begin{array}{c} \text{H} \\   \\ \text{H}-\text{N}-\text{H} \\   \\ \text{H} \end{array}$
Shape: <u>linear</u>	Shape: <u>linear</u>	Shape: <u>linear</u>	Shape: <u>trigonal pyramidal</u>
$\text{CH}_4$ VE's = <u>8</u> BE's = <u>8</u> RE's = <u>0</u> $\begin{array}{c} \text{H} \\   \\ \text{H}-\text{C}-\text{H} \\   \\ \text{H} \end{array}$	$\text{C}_2\text{H}_6$ VE's = <u>14</u> BE's = <u>14</u> RE's = <u>0</u> $\begin{array}{c} \text{H} \quad \text{H} \\   \quad   \\ \text{H}-\text{C}-\text{C}-\text{H} \\   \quad   \\ \text{H} \quad \text{H} \end{array}$	$\text{AlH}_3$ VE's = <u>6</u> BE's = <u>6</u> RE's = <u>0</u> $\begin{array}{c} \text{H}-\text{Al}-\text{H} \\   \\ \text{H} \end{array}$	$\text{BH}_3$ VE's = <u>6</u> BE's = <u>6</u> RE's = <u>0</u> $\begin{array}{c} \text{H} \\   \\ \text{H}-\text{B}-\text{H} \end{array}$
Shape: <u>tetrahedral</u>	Shape: <u>tetrahedral (both)</u>	Shape: <u>trigonal planar</u>	Shape: <u>trigonal planar</u>
$\text{C}_2\text{H}_4$ VE's = <u>12</u> BE's = <u>12</u> RE's = <u>0</u> $\begin{array}{c} \text{H} \quad \text{H} \\ \diagdown \quad / \\ \text{C}=\text{C} \\ / \quad \diagdown \\ \text{H} \quad \text{H} \end{array}$	$\text{C}_2\text{F}_4$ VE's = <u>36</u> BE's = <u>12</u> RE's = <u>24</u> $\begin{array}{c} \text{:}\ddot{\text{F}}\text{:} \quad \text{:}\ddot{\text{F}}\text{:} \\ \diagdown \quad / \\ \text{C}=\text{C} \\ / \quad \diagdown \\ \text{:}\ddot{\text{F}}\text{:} \quad \text{:}\ddot{\text{F}}\text{:} \end{array}$	$\text{CO}$ VE's = <u>10</u> BE's = <u>6</u> RE's = <u>4</u> $\text{:}\ddot{\text{C}}\equiv\ddot{\text{O}}\text{:}$	$\text{O}_2$ VE's = <u>12</u> BE's = <u>4</u> RE's = <u>8</u> $\text{:}\ddot{\text{O}}=\ddot{\text{O}}\text{:}$
Shape around one C: <u>trigonal planar</u>	Shape around one C: <u>trigonal planar</u>	Shape: <u>linear</u>	Shape: <u>linear</u>

# Lewis Structure Worksheet Answer Key

**L Reisser**



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## **Table of Contents Lewis Structure Worksheet Answer Key**

1. Understanding the eBook Lewis Structure Worksheet Answer Key
  - The Rise of Digital Reading Lewis Structure Worksheet Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Lewis Structure Worksheet Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Lewis Structure Worksheet Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Lewis Structure Worksheet Answer Key
  - Personalized Recommendations
  - Lewis Structure Worksheet Answer Key User Reviews and Ratings
  - Lewis Structure Worksheet Answer Key and Bestseller Lists

5. Accessing Lewis Structure Worksheet Answer Key Free and Paid eBooks
  - Lewis Structure Worksheet Answer Key Public Domain eBooks
  - Lewis Structure Worksheet Answer Key eBook Subscription Services
  - Lewis Structure Worksheet Answer Key Budget-Friendly Options
6. Navigating Lewis Structure Worksheet Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Lewis Structure Worksheet Answer Key Compatibility with Devices
  - Lewis Structure Worksheet Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Lewis Structure Worksheet Answer Key
  - Highlighting and Note-Taking Lewis Structure Worksheet Answer Key
  - Interactive Elements Lewis Structure Worksheet Answer Key
8. Staying Engaged with Lewis Structure Worksheet Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Lewis Structure Worksheet Answer Key
9. Balancing eBooks and Physical Books Lewis Structure Worksheet Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Lewis Structure Worksheet Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Lewis Structure Worksheet Answer Key
  - Setting Reading Goals Lewis Structure Worksheet Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Lewis Structure Worksheet Answer Key
  - Fact-Checking eBook Content of Lewis Structure Worksheet Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- 
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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