

Gas Laws Study Guide

Write the Law, Formula, Symbols, Units, and a sentence describing the meaning of the symbols and units.

1. A sample of gas is at 10.0 kPa and 20.0°C.

2. The pressure of a gas is 100.0 kPa and the volume is 2.00 L.

3. A sample of gas is at 10.0 kPa and 20.0°C.

4. A sample of gas is at 10.0 kPa and 20.0°C.

5. A sample of gas is at 10.0 kPa and 20.0°C.

6. A sample of gas is at 10.0 kPa and 20.0°C.

Name _____ Date _____

Gas Laws Study Guide

Standard Temperature = _____ °C = _____ K

Standard Pressure = _____ atm = _____ mmHg = _____ kPa = _____ mm Hg

Standard Volume = _____ L = _____ mL

Write the Formula for the Gas Law:

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

Combined Gas Law

$$P_{\text{Total}} = P_1 + P_2 + P_3$$

Charles' Law

$$P_1 V_1 = P_2 V_2$$

Boyle's Law

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

Gay-Lussac's Law

$$P_1 T_1 = P_2 T_2$$

Gay-Lussac's Law

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

Gay-Lussac's Law

Complete the following statements by filling in the blanks & writing the correct units.

As Pressure increases Volume decreases (increases) given that _____ is constant meaning that they have an inverse relationship.

As Volume increases Temperature increases (decreases) given that _____ is constant meaning that they have a direct inverse relationship.

As Pressure increases Temperature increases (decreases) given that _____ is constant meaning that they have a direct inverse relationship.

As the number of moles of gas in a container increases the Volume increases (decreases), the Pressure increases (decreases), and the Temperature increases (decreases).

Gas Laws Answer Study Guide Chemistry

Rachel Sandford



Gas Laws Answer Study Guide Chemistry:

Unveiling the Energy of Verbal Beauty: An Mental Sojourn through **Gas Laws Answer Study Guide Chemistry**

In a world inundated with screens and the cacophony of immediate conversation, the profound energy and psychological resonance of verbal beauty frequently disappear into obscurity, eclipsed by the regular assault of noise and distractions. Yet, situated within the musical pages of **Gas Laws Answer Study Guide Chemistry**, a fascinating function of literary splendor that pulses with fresh feelings, lies an unforgettable journey waiting to be embarked upon. Composed by way of a virtuoso wordsmith, this magical opus guides viewers on an emotional odyssey, softly exposing the latent potential and profound affect stuck within the complex internet of language. Within the heart-wrenching expanse with this evocative examination, we shall embark upon an introspective exploration of the book is central subjects, dissect its captivating publishing type, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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