

## Gas Laws Packet #2

### Ideal Gas Law Worksheet $PV = nRT$

Use the ideal gas law, " $PV = nRT$ ", and the universal gas constant  $R = 0.0821 \frac{L \cdot atm}{K \cdot mol}$  to solve the following problems:

If pressure is needed in kPa then convert by multiplying by  $101.3 kPa / 1 atm$  to get  
 $R = 8.31 \frac{L \cdot kPa}{K \cdot mole}$

- 1) If I have 4 moles of a gas at a pressure of 5.6 atm and a volume of 12 liters, what is the temperature?

**204.6 K**

- 2) If I have an unknown quantity of gas at a pressure of 1.2 atm, a volume of 31 liters, and a temperature of 37 °C, how many moles of gas do I have?

**1.26 mol**

- 3) If I contain 3 moles of gas in a container with a volume of 60 liters and at a temperature of 400 K, what is the pressure inside the container?

**1.64 atm**

- 4) If I have 7.7 moles of gas at a pressure of 0.09 atm and at a temperature of 56 °C, what is the volume of the container that the gas is in?

**2311 L**

- 5) If I have 17 moles of gas at a temperature of 67 °C, and a volume of 33.39 liters, what is the pressure of the gas?

**5.4 atm or 540 kPa**

- 6) If I have an unknown quantity of gas at a pressure of 0.5 atm, a volume of 25 liters, and a temperature of 300 K, how many moles of gas do I have?

**.5 mol**

# Gas Laws Packet Answers

**Anthony S. Fauci**



## Gas Laws Packet Answers:

## Unveiling the Power of Verbal Beauty: An Emotional Sojourn through **Gas Laws Packet Answers**

In a global inundated with displays and the cacophony of fast communication, the profound power and mental resonance of verbal art frequently diminish into obscurity, eclipsed by the regular onslaught of sound and distractions. However, located within the musical pages of **Gas Laws Packet Answers**, a charming perform of fictional splendor that impulses with organic thoughts, lies an unique journey waiting to be embarked upon. Published with a virtuoso wordsmith, that mesmerizing opus books readers on an emotional odyssey, lightly revealing the latent possible and profound influence embedded within the elaborate web of language. Within the heart-wrenching expanse of this evocative analysis, we shall embark upon an introspective exploration of the book is key styles, dissect its fascinating writing fashion, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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