

## Answers to the Gases Review Sheet

1. 3.21 mm Hg = 0.321 cm Hg = 0.00422 atm
2. 6.9 kPa = 6900 Pa = 52 mm Hg
3. 436 mm Hg = 0.574 atm = 58.1 kPa
4. 2.9 kPa = 22 torr = 22 mm Hg
5. A sample of air has a volume of 2.25 cm<sup>3</sup> when its temperature is 298 K. If the temperature is increased to 373 K without changing the pressure, what is the new volume of the sample in air?

$$\begin{aligned} V_1 &= 2.25 \text{ cm}^3 & \frac{V_1}{T_1} &= \frac{V_2}{T_2} & V_2 &= \frac{V_1 T_2}{T_1} \\ T_1 &= 298 \text{ K} \\ T_2 &= 373 \text{ K} \\ V_2 &= x \end{aligned}$$

*Charles's*

$$V_2 = \frac{2.25 \text{ cm}^3 (373 \text{ K})}{298 \text{ K}} = \mathbf{2.82 \text{ cm}^3}$$

6. As the temperature of a sample of nitrogen increases from 273 K, the volume of the sample changes from 275 mL to 325 mL. Assuming the pressure is constant, what is the final temperature of the gas?

$$\begin{aligned} V_1 &= 275 \text{ mL} & \frac{V_1}{T_1} &= \frac{V_2}{T_2} & T_2 &= \frac{T_1 V_2}{V_1} \\ T_1 &= 273 \text{ K} \\ T_2 &= x \\ V_2 &= 325 \text{ mL} \end{aligned}$$

*Charles's*

$$T_2 = \frac{325 \text{ mL} (273 \text{ K})}{275 \text{ mL}} = \mathbf{323 \text{ K}}$$

7. The pressure of a 5.71 L sample of neon gas is 23.4 kPa. Calculate the new pressure when the volume becomes 3.40 L.

$$\begin{aligned} V_1 &= 5.71 \text{ L} & P_1 V_1 &= P_2 V_2 & P_2 &= \frac{P_1 V_1}{V_2} \\ P_1 &= 23.4 \text{ kPa} \\ P_2 &= x \end{aligned}$$

$$P_2 = \frac{5.71 \text{ L} (23.4 \text{ kPa})}{3.40 \text{ L}} = 39.3 \text{ kPa} = \mathbf{0.388 \text{ atm}}$$

*Boyle's*

# Gases And Pressure Section Review Answers

**LP Steffe**



**Gases And Pressure Section Review Answers:**

## Reviewing **Gases And Pressure Section Review Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Gases And Pressure Section Review Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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