

NRG: Chemicals in Industry

Question: How do you determine the correct IUPAC name for a chemical? What are the conventions and guidelines that you follow in nomenclature?

How to Name an Ionic Compound

1. Identify the cation and anion.

Cation:	Na^+
Anion:	Cl^-
Formula:	NaCl
Name:	Sodium chloride

2. Name the cation first, then the anion.

Example: Sodium chloride

Formula: NaCl

How to Name a Molecular Compound

1. Identify the number of atoms of each element in the molecule.
 Example: CO_2 (Carbon dioxide)

2. Name the elements in order of increasing electronegativity.

Example: Carbon dioxide

3. Use prefixes to indicate the number of atoms of each element.

Example: Carbon dioxide

4. The second element's name ends in -ide.

Example: Carbon dioxide

5. The first element's name does not end in -ide.

Example: Carbon dioxide

6. The name of the compound is the sum of the names of the elements.

Example: Carbon dioxide

7. The name of the compound is the sum of the names of the elements.

Example: Carbon dioxide

8. The name of the compound is the sum of the names of the elements.

Example: Carbon dioxide

9. The name of the compound is the sum of the names of the elements.

Example: Carbon dioxide

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Example: Carbon dioxide

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Example: Carbon dioxide

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Example: Carbon dioxide

14. The name of the compound is the sum of the names of the elements.

Example: Carbon dioxide

15. The name of the compound is the sum of the names of the elements.

Example: Carbon dioxide

NRG

Question: How do you determine the correct IUPAC name for a chemical? What are the conventions and guidelines that you follow in nomenclature?

How to Name an Ionic Compound

1. Identify the cation and anion.
 Example: NaCl (Sodium chloride)

2. Name the cation first, then the anion.
 Example: Sodium chloride

3. Use the appropriate prefix to indicate the number of atoms of each element.
 Example: Sodium chloride

4. The second element's name ends in -ide.
 Example: Sodium chloride

5. The first element's name does not end in -ide.
 Example: Sodium chloride

6. The name of the compound is the sum of the names of the elements.
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9. The name of the compound is the sum of the names of the elements.
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10. The name of the compound is the sum of the names of the elements.
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11. The name of the compound is the sum of the names of the elements.
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12. The name of the compound is the sum of the names of the elements.
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13. The name of the compound is the sum of the names of the elements.
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14. The name of the compound is the sum of the names of the elements.
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15. The name of the compound is the sum of the names of the elements.
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18. The name of the compound is the sum of the names of the elements.
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 Example: Sodium chloride

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1. Understanding the eBook Icivics Answer Keys
 - The Rise of Digital Reading Icivics Answer Keys
 - Advantages of eBooks Over Traditional Books
2. Identifying Icivics Answer Keys
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Icivics Answer Keys
 - User-Friendly Interface
4. Exploring eBook Recommendations from Icivics Answer Keys
 - Personalized Recommendations
 - Icivics Answer Keys User Reviews and Ratings
 - Icivics Answer Keys and Bestseller Lists
5. Accessing Icivics Answer Keys Free and Paid eBooks
 - Icivics Answer Keys Public Domain eBooks
 - Icivics Answer Keys eBook Subscription Services
 - Icivics Answer Keys Budget-Friendly Options

6. Navigating Icivics Answer Keys eBook Formats
 - ePub, PDF, MOBI, and More
 - Icivics Answer Keys Compatibility with Devices
 - Icivics Answer Keys Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Icivics Answer Keys
 - Highlighting and Note-Taking Icivics Answer Keys
 - Interactive Elements Icivics Answer Keys
8. Staying Engaged with Icivics Answer Keys
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Icivics Answer Keys
9. Balancing eBooks and Physical Books Icivics Answer Keys
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Icivics Answer Keys
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Icivics Answer Keys
 - Setting Reading Goals Icivics Answer Keys
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Icivics Answer Keys
 - Fact-Checking eBook Content of Icivics Answer Keys
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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