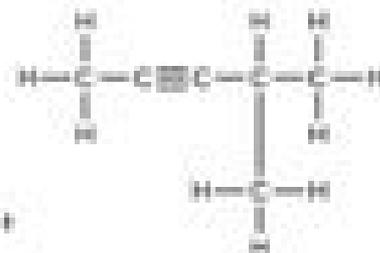
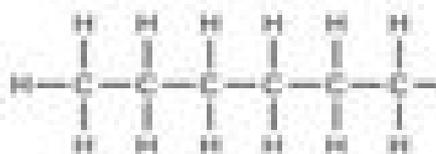


TOPIC 1 -

Hydrocarbons



Organic compounds containing only hydrogen and carbon atoms.

Key Facts

- Hydrocarbons may have one or more branched chains or cyclic structures.
- The hydrogen and carbon atoms have similar electronegativities (2.1 and 2.5) making them have low solubility in polar solvents.
- Hydrocarbons have the weakest intermolecular forces (London dispersion forces).
- Weak intermolecular forces → low boiling and melting points.
- 3 main groups of hydrocarbons:
 - Alkanes ⇒ single carbon-carbon bonds (saturated hydrocarbons)
 - Alkenes ⇒ contain at least one carbon-carbon double bond (unsaturated hydrocarbons)
 - Alkynes ⇒ contain at least one carbon-carbon triple bond (unsaturated)

3 Main Groups of Hydrocarbons

Alkanes

- General formula = $\text{C}_n\text{H}_{2n+2}$
- Hydrocarbons with single

Alkenes

- General formula = C_nH_{2n}
- Hydrocarbons with at least

Alkynes

- General formula = $\text{C}_n\text{H}_{2n-2}$
- Hydrocarbons with at least

Hydrocarbons Workbook Answer Key

B Lingard



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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Hydrocarbons Workbook Answer Key**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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