

Honors Chemistry – Unit 3 Review

Complete each of the following problems before or after a period of instruction and before the next class. Answers should be written on separate sheets of paper. Do not write your name on these sheets.

Review

Write a story, incorporating each of the following ideas, describing a change in the state of matter. Which substances/elements or mixtures change phase, and give the names of the substances and state changes. What conditions (temperature, pressure, etc.) are needed to bring about the change? Use particles, molecules, atoms, ions, etc. to describe the change. Be sure to include the names of the substances involved. Be sure to include the names of the substances involved.

1. Explain why the condensation of water vapor is a physical change and not a chemical change.

2. Explain why the combustion of hydrogen gas is a chemical change and not a physical change.

3. Explain how the formation of a solution is a physical change and not a chemical change. Give an example.

Intermolecular Forces – Pressure Concepts

Write a story, incorporating all of the following ideas, describing a change in the state of matter. Which substances/elements or mixtures change phase, and give the names of the substances and state changes. What conditions (temperature, pressure, etc.) are needed to bring about the change? Use particles, molecules, atoms, ions, etc. to describe the change. Be sure to include the names of the substances involved.

4. Explain the relationship between pressure and volume.

5. Explain how pressure affects the boiling point of a liquid.

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Marquis Who's Who



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