

element	# valence e ⁻	# e ⁻ needed	gained / lost	ion symbol	name
B	3 <small>(group 13)</small>	3	lost <small>(+3 valence)</small>	B⁺³	boron cation
F	7 <small>(group 17)</small>	1	gained <small>(-1 valence)</small>	F⁻¹	fluoride anion
Li	1	1	lost	Li⁺¹	lithium cation
N	5	3	gained	N⁻³	nitride anion
O	6	2	gained	O⁻²	oxide anion
Ca	2	2	lost	Ca⁺²	calcium cation
Br	7	1	gained	Br⁻¹	bromide anion
S	6	2	gained	S⁻²	sulfide anion
Cl	7	1	gained	Cl⁻¹	chloride anion
Ba	2	2	lost	Ba⁺²	barium cation
K	1	1	lost	K⁺¹	potassium cation
Mg	2	2	lost	Mg⁺²	magnesium cation
Se	6	2	gained	Se⁻²	selenide anion
C	4	4	gained	C⁻⁴	carbide anion
Al	3	3	lost	Al⁺³	aluminum cation
Sn	4	4	lost	Sn⁺⁴	tin (IV) cation or tin cation

Monatomic Ions Packet Answers

M Walker



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