

Skills Worksheet

**Concept Review Chapter 5: Section 1**

**Section: Simple Ions**

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

- |    |    |       |        |
|----|----|-------|--------|
| 10 | 11 | ion   | octet  |
| 12 | 13 | anion | cation |

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?  
They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

# Holt Chemistry Chapter Concept Review Answers

**Thomas Griffiths**



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## **Table of Contents Holt Chemistry Chapter Concept Review Answers**

1. Understanding the eBook Holt Chemistry Chapter Concept Review Answers
  - The Rise of Digital Reading Holt Chemistry Chapter Concept Review Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Holt Chemistry Chapter Concept Review Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Holt Chemistry Chapter Concept Review Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Chemistry Chapter Concept Review Answers
  - Personalized Recommendations
  - Holt Chemistry Chapter Concept Review Answers User Reviews and Ratings
  - Holt Chemistry Chapter Concept Review Answers and Bestseller Lists
5. Accessing Holt Chemistry Chapter Concept Review Answers Free and Paid eBooks
  - Holt Chemistry Chapter Concept Review Answers Public Domain eBooks
  - Holt Chemistry Chapter Concept Review Answers eBook Subscription Services
  - Holt Chemistry Chapter Concept Review Answers Budget-Friendly Options
6. Navigating Holt Chemistry Chapter Concept Review Answers eBook Formats

- ePub, PDF, MOBI, and More
  - Holt Chemistry Chapter Concept Review Answers Compatibility with Devices
  - Holt Chemistry Chapter Concept Review Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Holt Chemistry Chapter Concept Review Answers
  - Highlighting and Note-Taking Holt Chemistry Chapter Concept Review Answers
  - Interactive Elements Holt Chemistry Chapter Concept Review Answers
8. Staying Engaged with Holt Chemistry Chapter Concept Review Answers
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Holt Chemistry Chapter Concept Review Answers
9. Balancing eBooks and Physical Books Holt Chemistry Chapter Concept Review Answers
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Holt Chemistry Chapter Concept Review Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Holt Chemistry Chapter Concept Review Answers
- Setting Reading Goals Holt Chemistry Chapter Concept Review Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Holt Chemistry Chapter Concept Review Answers
- Fact-Checking eBook Content of Holt Chemistry Chapter Concept Review Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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