

Final study guide question and answers

What kind of equilibrium exists and under what conditions? **Answer:** chemical equilibrium
Equilibrium is achieved for any and all closed systems containing homogeneous matter.
Equilibrium is

Microscopic: all chemical substances present, and any chemical reaction which might occur, is occurring, at identical rates. **Answer:** chemical equilibrium

What makes a law? **Answer:** The degree of homogeneity that is achieved. **Answer:** chemical equilibrium that is reached

What points in a system's phase diagram are important? **Answer:**

Triple point: where all three phases coexist
Critical point: where liquid and gas phases coexist

Normal boiling point: where liquid and gas phases coexist

What is the boiling point of a liquid at a given pressure? **Answer:** the temperature at which the liquid and gas phases coexist

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Rex Miller, Mark Miller



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