

1. Read the given statements and select the correct option.

Statement 1 : The formula connecting u, v and f for a spherical mirror is valid only for mirrors whose sizes are very small compared to their radii of curvature.

Statement 2 : Laws of reflection are strictly valid for plane surfaces, but not for large spherical surfaces.

- (A) Both statements 1 and 2 are true and statement 2 is the correct explanation of statement 1.
(B) Both statements 1 and 2 are true but statement 2 is not the correct explanation of statement 1.
(C) Statement 1 is true but statement 2 is false.
(D) Statement 1 is false but statement 2 is true.

2. A boy throws a table tennis ball of mass 20 g upwards with a velocity of $u_0 = 10$ m/s at an angle θ_0 with the vertical. The wind imparts a horizontal force of 0.08 N, so that the ball returns to the starting point. Then, the angle θ_0 must be such that, $\tan \theta_0$ is

- (A) 0.2 (B) 0.4 (C) 2.5 (D) 1.2

3. A weight is attached to the free end of a sonometer wire. It gives resonance at a length 40 cm when it is resonanced with a tuning fork of frequency 51 Hz. The weight is then immersed

wholly in water, the resonant length is reduced to 30 cm. The relative density in which weight suspended is

- (A) 16/9 (B) 16/7 (C) 16/5 (D) 16/3

4. Hydrogen sulphide (H_2S) contains 94.11% sulphur, water (H_2O) contains 11.11% hydrogen and sulphur dioxide (SO_2) contains 50% oxygen. Find the ratio of all given elements. After your calculations which law has been verified?

- (A) Law of multiple proportions
(B) Law of reciprocal proportions
(C) Law of constant composition
(D) Law of combining volumes

5. Which of the following represents the given mode of hybridisation, $sp^2-sp^2-sp-sp$ from left to right?

- (A) $\text{CH}_2=\text{CH}-\text{C}\equiv\text{N}$ (B) $\text{CH}\equiv\text{C}-\text{C}\equiv\text{CH}$
(C) $\text{CH}_2=\text{C}=\text{C}=\text{CH}_2$ (D) 

6. In the reaction,



when 1 mole of ammonia and 1 mole of O_2 are made to react to completion

- (A) 1.0 mole of H_2O is produced
(B) 2.0 moles of NO will be produced
(C) All the oxygen will be consumed
(D) All the ammonia will be consumed.

National Science Olympiad Sample Papers

LL Leslie



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Table of Contents National Science Olympiad Sample Papers

1. Understanding the eBook National Science Olympiad Sample Papers
 - The Rise of Digital Reading National Science Olympiad Sample Papers
 - Advantages of eBooks Over Traditional Books
2. Identifying National Science Olympiad Sample Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an National Science Olympiad Sample Papers
 - User-Friendly Interface
4. Exploring eBook Recommendations from National Science Olympiad Sample Papers
 - Personalized Recommendations
 - National Science Olympiad Sample Papers User Reviews and Ratings
 - National Science Olympiad Sample Papers and Bestseller Lists
5. Accessing National Science Olympiad Sample Papers Free and Paid eBooks
 - National Science Olympiad Sample Papers Public Domain eBooks
 - National Science Olympiad Sample Papers eBook Subscription Services
 - National Science Olympiad Sample Papers Budget-Friendly Options

6. Navigating National Science Olympiad Sample Papers eBook Formats
 - ePub, PDF, MOBI, and More
 - National Science Olympiad Sample Papers Compatibility with Devices
 - National Science Olympiad Sample Papers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of National Science Olympiad Sample Papers
 - Highlighting and Note-Taking National Science Olympiad Sample Papers
 - Interactive Elements National Science Olympiad Sample Papers
8. Staying Engaged with National Science Olympiad Sample Papers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers National Science Olympiad Sample Papers
9. Balancing eBooks and Physical Books National Science Olympiad Sample Papers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection National Science Olympiad Sample Papers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine National Science Olympiad Sample Papers
 - Setting Reading Goals National Science Olympiad Sample Papers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of National Science Olympiad Sample Papers
 - Fact-Checking eBook Content of National Science Olympiad Sample Papers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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