



# THE NATURE OF GASES

## Section Review

### Objectives

1. describe the relationship between Boyle's and Charles's laws
2. describe Graham's law of effusion
3. derive the combined gas law and apply it to solve problems

### Vocabulary

- |                     |                       |                             |
|---------------------|-----------------------|-----------------------------|
| a. Boyle's law      | b. Charles's law      | c. combined gas law         |
| d. Boyle's constant | e. Charles's constant | f. Graham's law of effusion |
| g. Boyle's equation | h. Charles's equation |                             |

### Part A: Completion

Fill in the missing terms in each sentence. Write your answers on the lines provided.

Boyle's law states that the \_\_\_\_\_ of a gas is inversely proportional to its volume when the temperature is held constant. The \_\_\_\_\_ of a gas is the product of its pressure and volume. \_\_\_\_\_ states that the volume of a gas is directly proportional to its absolute temperature when the \_\_\_\_\_ is held constant. \_\_\_\_\_ states that the rate of effusion of a gas is inversely proportional to the square root of its molecular weight. \_\_\_\_\_ states that the pressure of a gas is directly proportional to its absolute temperature when its volume and the number of moles of gas are held constant. \_\_\_\_\_ states that the rate of effusion of a gas is inversely proportional to the square root of its molecular weight.

1	_____
2	_____
3	_____
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5	_____
6	_____
7	_____
8	_____
9	_____
10	_____

### Part B: True-False

Write **T** for true and **F** for false on the lines provided.

- \_\_\_\_\_ 11. Boyle's law states that the volume of a gas is directly proportional to its pressure when the temperature is held constant.
- \_\_\_\_\_ 12. Charles's law states that the volume of a gas is directly proportional to its absolute temperature when the pressure is held constant.

# Nature Of Gases Section Review Answer Key

**S Nieto**



**Nature Of Gases Section Review Answer Key:**

## Reviewing **Nature Of Gases Section Review Answer Key**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Nature Of Gases Section Review Answer Key**," an enthralling opus penned by a highly acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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