

1. For **each** of the following sets of sublevels, determine which has the lowest energy, and which has the highest energy.

(a) 2s, 3s, 4s

(b) 4f, 4p, 4d

(c) 4d, 4f, 3d

(d) 6s, 3d, 5f

Ex for (a) Lowest 2s, highest 4s

2. Explain what  $6s^2$  means.

3. What is the maximum number of electrons that can be found in an s – sublevel? \_\_\_ p – sublevel? \_\_\_ d – sublevel? \_\_\_ f – sublevel? \_\_\_

4. Indicate which of these (there may be more than one) orbital designations is incorrect:

(a) 4s

(b) 3f

(c) 2d

(d) 3d

5. Draw energy level diagrams for

(a) sodium

(b) silicon

(c) beryllium

(d) cobalt

6. How many unpaired electrons are there in each of the atoms in question 5?

7. Give the electron configuration for:

(a) lithium

(b) oxygen

(c) selenium

(d) iodine

(e) silver

(f) arsenic

(g) indium

(h) potassium

8. Give the name and symbol for the atom whose electron configuration corresponds to each of the following electron configurations.

a)  $1s^2 2s^2 2p^6 3s^2 3p^5$

b)  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^3$

c)  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^6$

d)  $1s^2 2s^2 2p^6 3s^2 3p^2$

e)  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1$

9. Write the shorthand notation for the above electron configurations.

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**Léo Pomar**



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## **Table of Contents Electron Configuration Extension Questions**

1. Understanding the eBook Electron Configuration Extension Questions
  - The Rise of Digital Reading Electron Configuration Extension Questions
  - Advantages of eBooks Over Traditional Books
2. Identifying Electron Configuration Extension Questions
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Electron Configuration Extension Questions
  - User-Friendly Interface

4. Exploring eBook Recommendations from Electron Configuration Extension Questions
  - Personalized Recommendations
  - Electron Configuration Extension Questions User Reviews and Ratings
  - Electron Configuration Extension Questions and Bestseller Lists
5. Accessing Electron Configuration Extension Questions Free and Paid eBooks
  - Electron Configuration Extension Questions Public Domain eBooks
  - Electron Configuration Extension Questions eBook Subscription Services
  - Electron Configuration Extension Questions Budget-Friendly Options
6. Navigating Electron Configuration Extension Questions eBook Formats
  - ePub, PDF, MOBI, and More
  - Electron Configuration Extension Questions Compatibility with Devices
  - Electron Configuration Extension Questions Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Electron Configuration Extension Questions
  - Highlighting and Note-Taking Electron Configuration Extension Questions
  - Interactive Elements Electron Configuration Extension Questions
8. Staying Engaged with Electron Configuration Extension Questions
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Electron Configuration Extension Questions
9. Balancing eBooks and Physical Books Electron Configuration Extension Questions
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Electron Configuration Extension Questions
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Electron Configuration Extension Questions
  - Setting Reading Goals Electron Configuration Extension Questions
  - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Electron Configuration Extension Questions
  - Fact-Checking eBook Content of Electron Configuration Extension Questions
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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psychological approaches psychological inflexibility or avoidance of one's own cognitive and affective states at a cost to personal values may be a useful construct to understand these problems

body dysmorphic disorder johns hopkins medicine - Aug 03 2022

web body dysmorphic disorder bdd is a mental health problem if you have bdd you may be so upset about the appearance of your body that it gets in the way of your ability to live normally many of us have what we think are flaws in our appearance but if you have bdd your reaction to this flaw may become overwhelming